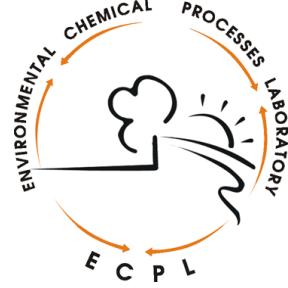




TM-meeting, 30-31 May 2011, ISPRA



TM4-ECPL modeling activities

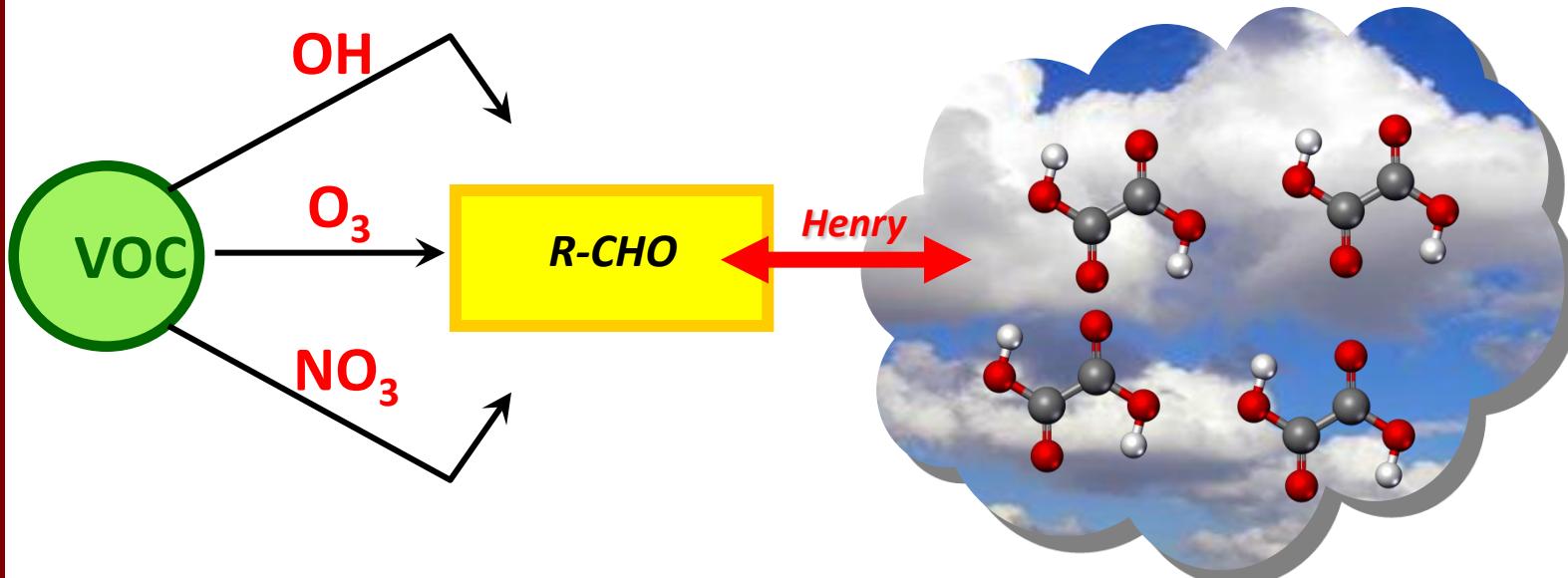
Stelios Myriokefalitakis, Nikos Daskalakis, Kanakidou Maria
Environmental Chemical Process Laboratory (ECPL),
Department of Chemistry, University of Crete

Kostas Tsigaridis
NASA, GISS & Uni Columbia, NY

- 1. Multiphase chemistry (Stelios)**
2. Interannual model evaluation & AEROCOM
(Nikos)
3. Other ongoing and future activities (Maria)

SOA formation via aqueous-phase chemistry in TM4-ECPL

1. VOC photo-oxidation in the gas-phase
2. Production of water-soluble organic compounds in the gas-phase (*e.g. aldehydes, organic acids*)
3. Phase transfer between the gas and the aqueous phase
4. Production of low volatile compounds in the aqueous-phase (*e.g. oxalic acid*)
5. Upon cloud evaporation new organic particulate matter is formed



The Revised Aqueous-Phase Chemical Scheme in TM4-ECPL

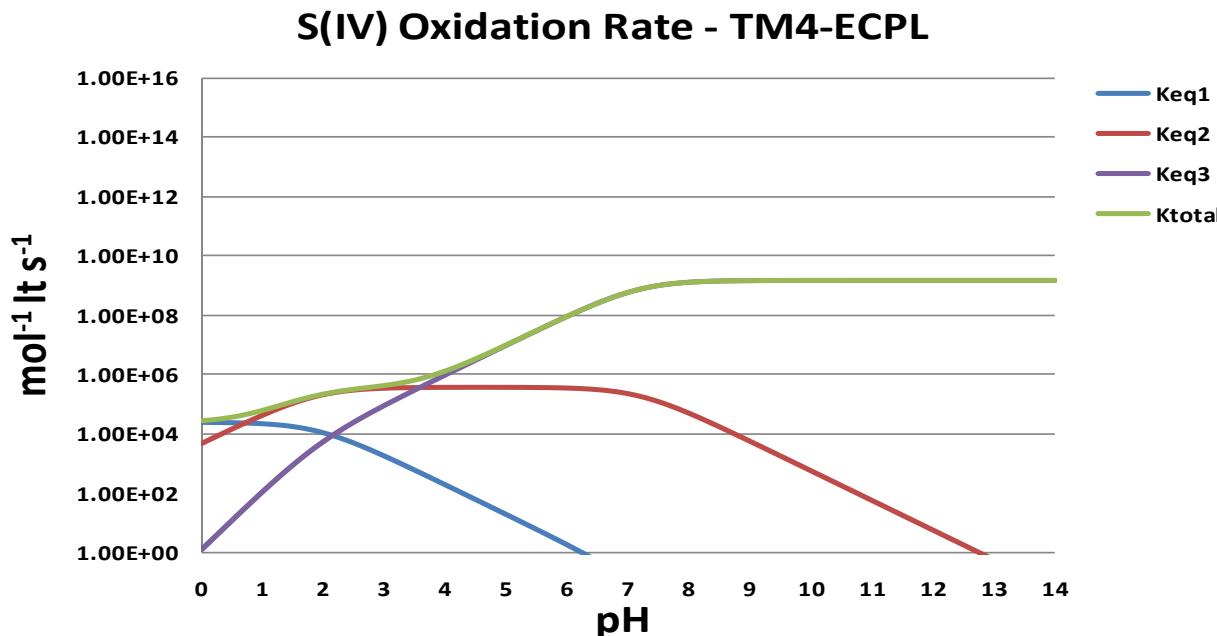
1. NO₃ Aqueous-phase Oxidation of Aldehydes and Organic Acids
2. Lower Henry Constant of OH radicals ($9 \cdot 10^3 \text{ mol L}^{-1} \text{ atm}^{-1}$ -> $30 \text{ mol L}^{-1} \text{ atm}^{-1}$)
3. Different cloud droplet radius over land (5 μm) and over oceans (10 μm)
4. Simultaneously S(IV) and Organic molecules aqueous-phase oxidation using EBI solver

pH Dependence of S(IV) oxidation by ozone

- TM4-ECPL does not take into account the different forms of the sulfuric acid ($\text{SO}_2 \cdot \text{H}_2\text{O}$, HSO_3^- , SO_3^{2-}). , in order to take into account the different forms and reaction rates, we use the mole fraction (ξ) of each acid as a function of pH. The total mass of S(IV) is calculated as $\text{S(IV)} = \text{SO}_2 \cdot \text{H}_2\text{O} + \text{HSO}_3^- + \text{SO}_3^{2-}$.

$$\xi_{\text{SO}_2 \cdot \text{H}_2\text{O}} = \frac{[\text{SO}_2 \cdot \text{H}_2\text{O}]}{[\text{S(IV)}]} = \left(1 + \frac{\text{Keq}_1}{[\text{H}^+]} + \frac{\text{Keq}_1 \text{Keq}_2}{[\text{H}^+]^2} \right)^{-1} \quad \xi_{\text{HSO}_3^-} = \frac{[\text{HSO}_3^-]}{[\text{S(IV)}]} = \left(1 + \frac{[\text{H}^+]}{\text{Keq}_1} + \frac{\text{Keq}_2}{[\text{H}^+]} \right)^{-1} \quad \xi_{\text{SO}_3^{2-}} = \frac{[\text{SO}_3^{2-}]}{[\text{S(IV)}]} = \left(1 + \frac{[\text{H}^+]}{\text{Keq}_2} + \frac{[\text{H}^+]^2}{\text{Keq}_1 \text{Keq}_2} \right)^{-1}$$

- For 25°C, the oxidation of $\text{SO}_2 \cdot \text{H}_2\text{O}$, HSO_3^- , SO_3^{2-} accounts a rate of $\text{Keq}_1 = 2.4 \cdot 10^4 \text{ mol}^{-1} \text{ lt s}^{-1}$, $\text{Keq}_2 = 3.7 \cdot 10^5 \text{ mol}^{-1} \text{ lt s}^{-1}$ and $\text{Keq}_3 = 1.5 \cdot 10^9 \text{ mol}^{-1} \text{ lt s}^{-1}$ proportionally (Seinfeld and Pandis, 1998).
- $K_{\text{total}}(\text{SIV}) = \text{Keq}_1 * \xi_{\text{SO}_2 \cdot \text{H}_2\text{O}} + \text{Keq}_2 * \xi_{\text{HSO}_3^-} + \text{Keq}_3 * \xi_{\text{SO}_3^{2-}}$

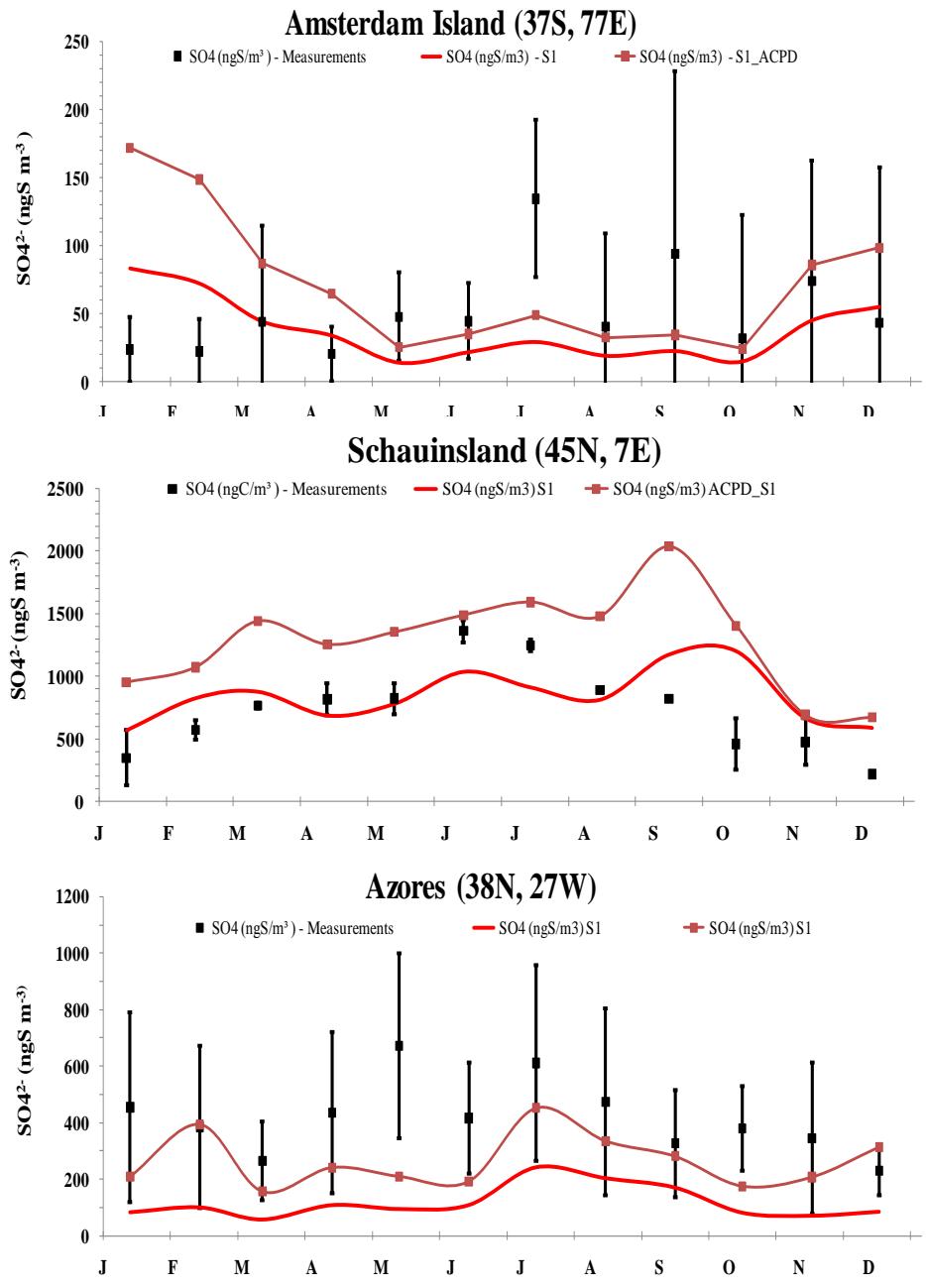
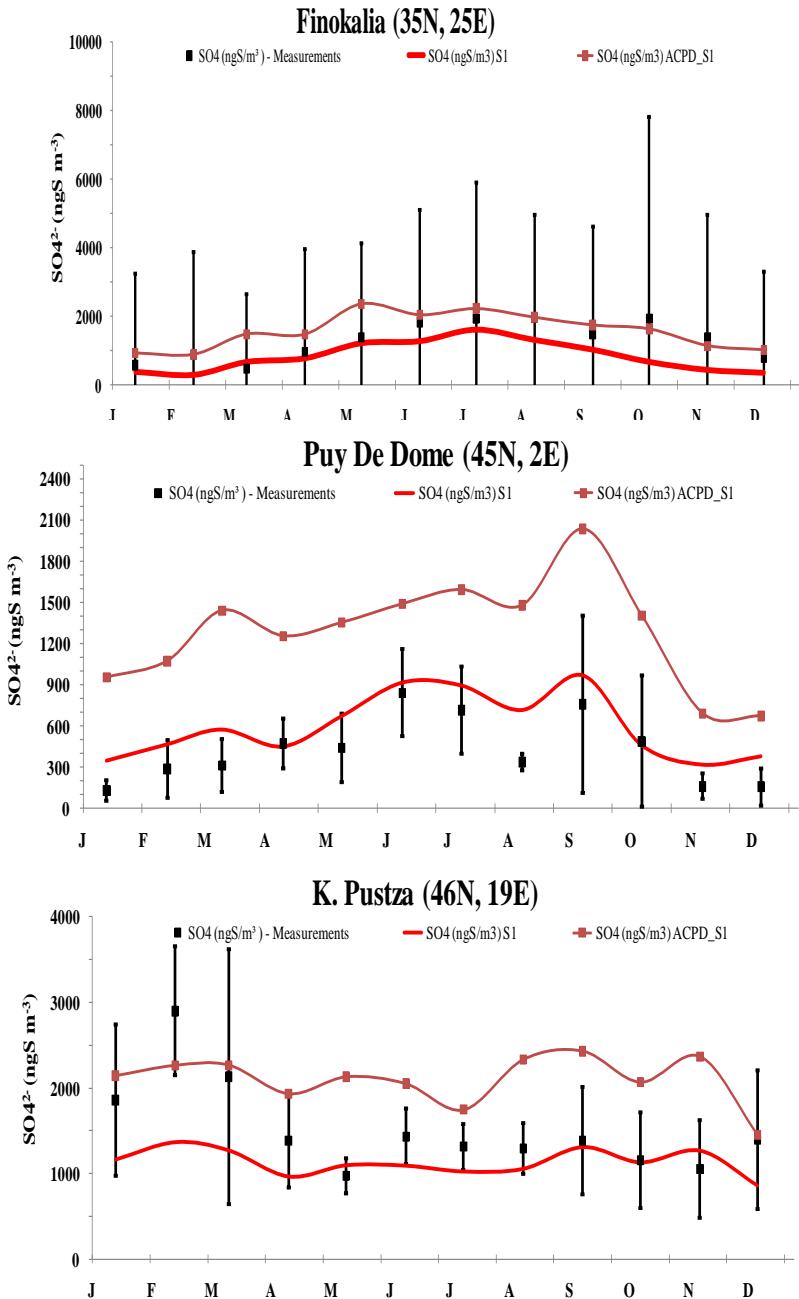


Inorganic Aqueous-Phase Chemistry

Aqueous Phase Reactions



SO_4^{2-} -validation in TM4-ECPL



Organic Aqueous-Phase Chemical Scheme – Aldehydes (C1-C3)

Aqueous Phase Reactions

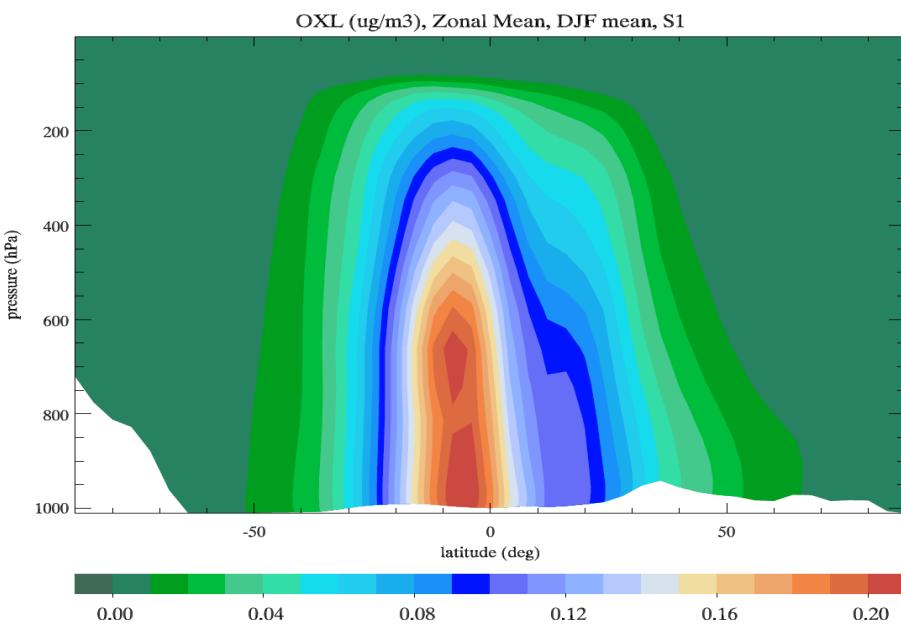
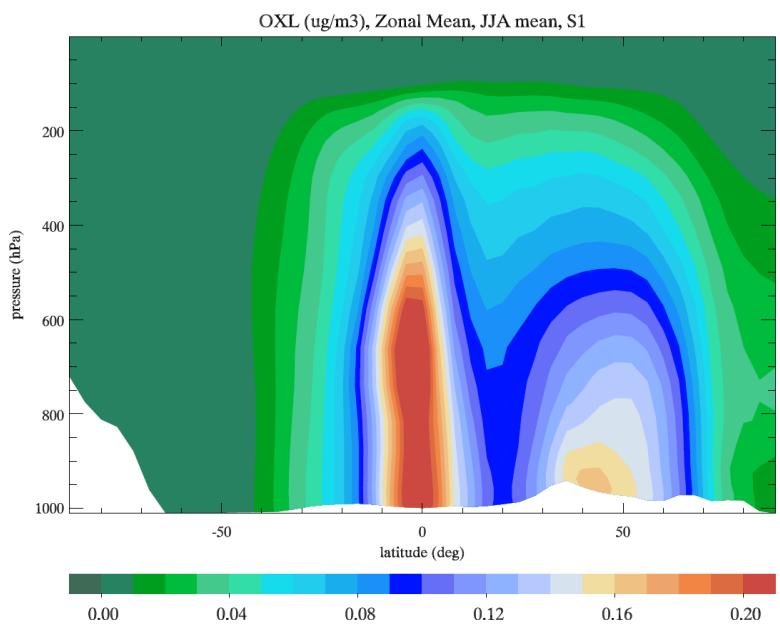
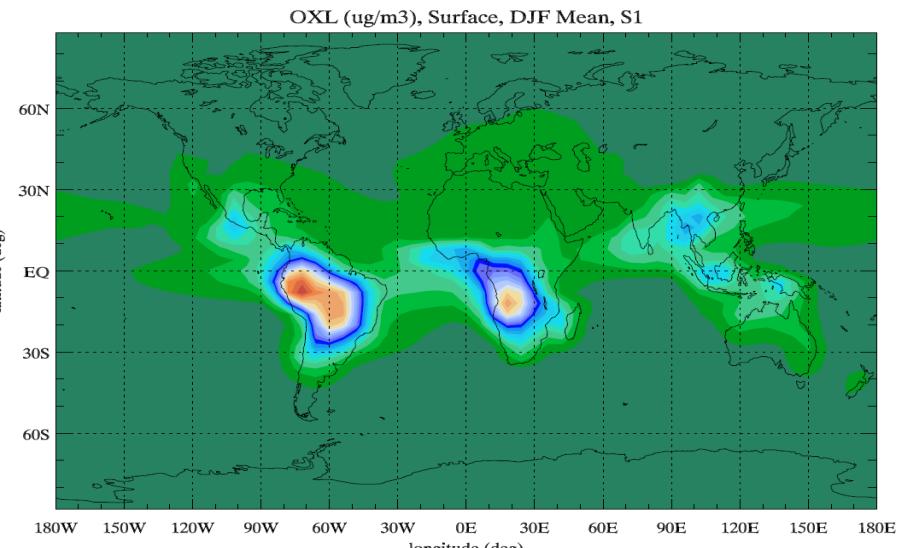
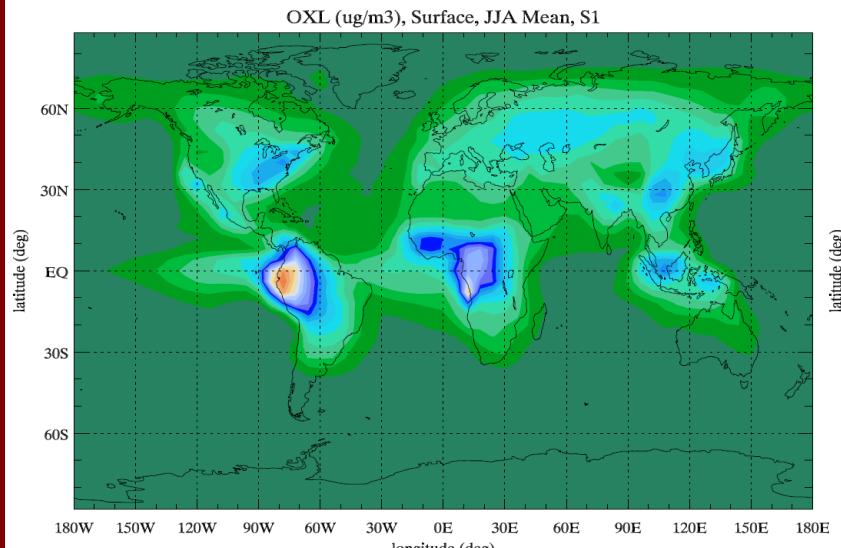
14	$\text{CH}_2(\text{OH})_2 + \text{OH} (+ \text{O}_2)$	\rightarrow	$\text{HCOOH} + \text{HO}_2 + \text{H}_2\text{O}$
15	$\text{CH}_2(\text{OH})_2 + \text{NO}_3 (+ \text{O}_2)$	\rightarrow	$\text{HCOOH} + \text{HO}_2 + \text{NO}_3^- + \text{H}^+$
16	$\text{GLYAL} + \text{OH} (+ \text{O}_2)$	\rightarrow	$\text{GLY} + \text{HO}_2$
17	$\text{GLYAL} + 2 \text{OH} (+ 2 \text{O}_2)$	\rightarrow	$\text{GLX} + 2\text{HO}_2 + 2 \text{H}_2\text{O}$
18	$\text{GLYAL} + \text{NO}_3 (+ \text{O}_2)$	\rightarrow	$\text{GLX} + \text{HO}_2 + \text{NO}_3^- + \text{H}^+$
19	$\text{GLYAL} + 2 \text{NO}_3 (+ \text{O}_2)$	\rightarrow	$\text{GLY} + 2 \text{NO}_3^- + 2 \text{H}^+ + \text{H}_2\text{O}$
20	$\text{GLY} + \text{OH} (+ \text{O}_2)$	\rightarrow	$\text{GLX} + \text{HO}_2 + \text{H}_2\text{O}$
21	$\text{GLY} + \text{OH}$	\rightarrow	$0.03\text{GLX} + 0.97\text{OXL} + \text{H}_2\text{O}$
22	$\text{GLY} + \text{NO}_3 (+ \text{O}_2)$	\rightarrow	$\text{GLX} + \text{HO}_2 + \text{NO}_3^- + \text{H}^+$
23	$\text{GLY} + h\nu/\text{OH}$ <i>(only in aerosol water)</i>	\rightarrow	$0.2\text{OXL} + 0.8\text{OLIGOMERIC-SOA}$
24	$\text{GLY} + \text{NH}_4^+$ <i>(only in aerosol water)</i>	\rightarrow	OLIGOMERIC-SOA
25	$\text{MGLY} + \text{OH} (+ \text{O}_2)$	\rightarrow	$0.92\text{PRV} + 0.08\text{GLX} + \text{HO}_2 + \text{H}_2\text{O}$
26	$\text{MGLY} + \text{NO}_3 (+ \text{O}_2)$	\rightarrow	$0.92\text{PRV} + 0.08\text{GLX} + \text{HO}_2 + \text{NO}_3^- + \text{H}^+$

Organic Aqueous-Phase Chemical Scheme – Organic Acids

Aqueous Phase Reactions

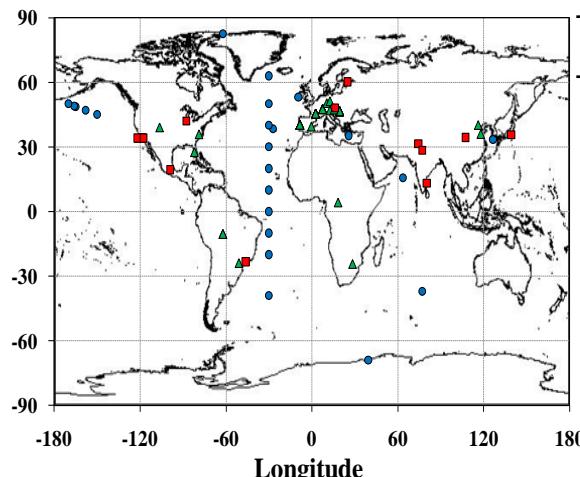
HCOOH + OH (+ O ₂)	→	CO ₂ + HO ₂ + H ₂ O
HCOO ⁻ + OH (+ O ₂)	→	CO ₂ + H ₂ O (+ O ₂ ⁻)
HCOOH + NO ₃ (+ O ₂)	→	CO ₂ + NO ₃ ⁻ + 2H ⁺ (+ O ₂ ⁻)
HCOO ⁻ + NO ₃ (+ O ₂)	→	CO ₂ + NO ₃ ⁻ + H ⁺ (+ O ₂ ⁻)
CH ₃ COOH + OH (+ O ₂)	→	0.85GLX + 0.15CH ₂ (OH) ₂
CH ₃ COO ⁻ + OH (+ O ₂)	→	0.85GLX ⁻ + 0.15CH ₂ (OH) ₂
CH ₃ COOH + NO ₃ (+ O ₂)	→	0.85GLX + 0.15CH ₂ (OH) ₂ + NO ₃ ⁻ + H ⁺
CH ₃ COO ⁻ + NO ₃ (+ O ₂)	→	0.85GLX ⁻ + 0.15CH ₂ (OH) ₂ + NO ₃ ⁻ + H ⁺
PRV + OH (+ O ₂)	→	CH ₃ COOH + HO ₂ + CO ₂
PRV ⁻ + OH	→	CH ₃ COO ⁻ + HO ₂ + CO ₂
PRV + NO ₃ (+ O ₂ + H ₂ O)	→	CH ₃ COOH + CO ₂ + HO ₂ + NO ₃ ⁻ + H ⁺
PRV ⁻ + NO ₃ (+ O ₂ + H ₂ O)	→	CH ₃ COO ⁻ + CO ₂ + HO ₂ + NO ₃ ⁻ + H ⁺
GLX + OH (+ O ₂)	→	OXL + HO ₂ + H ₂ O
GLX ⁻ + OH (+O ₂)	→	OXL ⁻ + HO ₂ + H ₂ O
GLX + NO ₃ (+ O ₂)	→	OXL + HO ₂ + NO ₃ ⁻ + H ⁺
GLX ⁻ + NO ₃ (+ O ₂)	→	OXL ⁻ + HO ₂ + NO ₃ ⁻ + H ⁺
OXL + 2OH	→	2CO ₂ + 2H ₂ O
OXL ⁻ + OH (+ O ₂)	→	2CO ₂ + H ₂ O (+ O ₂ ⁻)
OXL ²⁻ + OH (+ O ₂)	→	2CO ₂ + HO ⁻ (+ O ₂ ⁻)
OXL + 2NO ₃	→	2CO ₂ + 2NO ₃ ⁻ + 2H ⁺
OXL ⁻ + NO ₃ (+ O ₂)	→	2CO ₂ + NO ₃ ⁻ + H ⁺ (+ O ₂ ⁻)
OXL ²⁻ + NO ₃ (+ O ₂)	→	2CO ₂ + NO ₃ ⁻ (+ O ₂ ⁻)

OXL Calculated Distributions

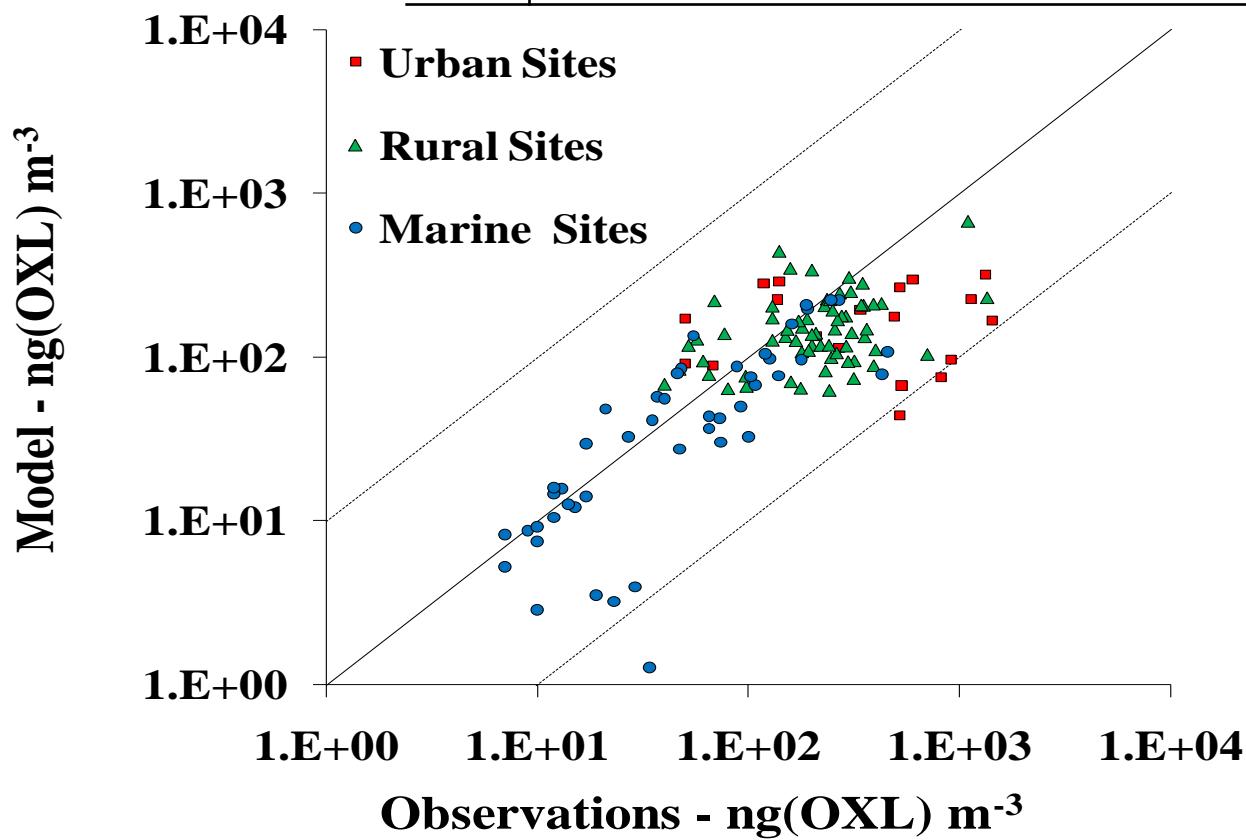


Global OXL Validations

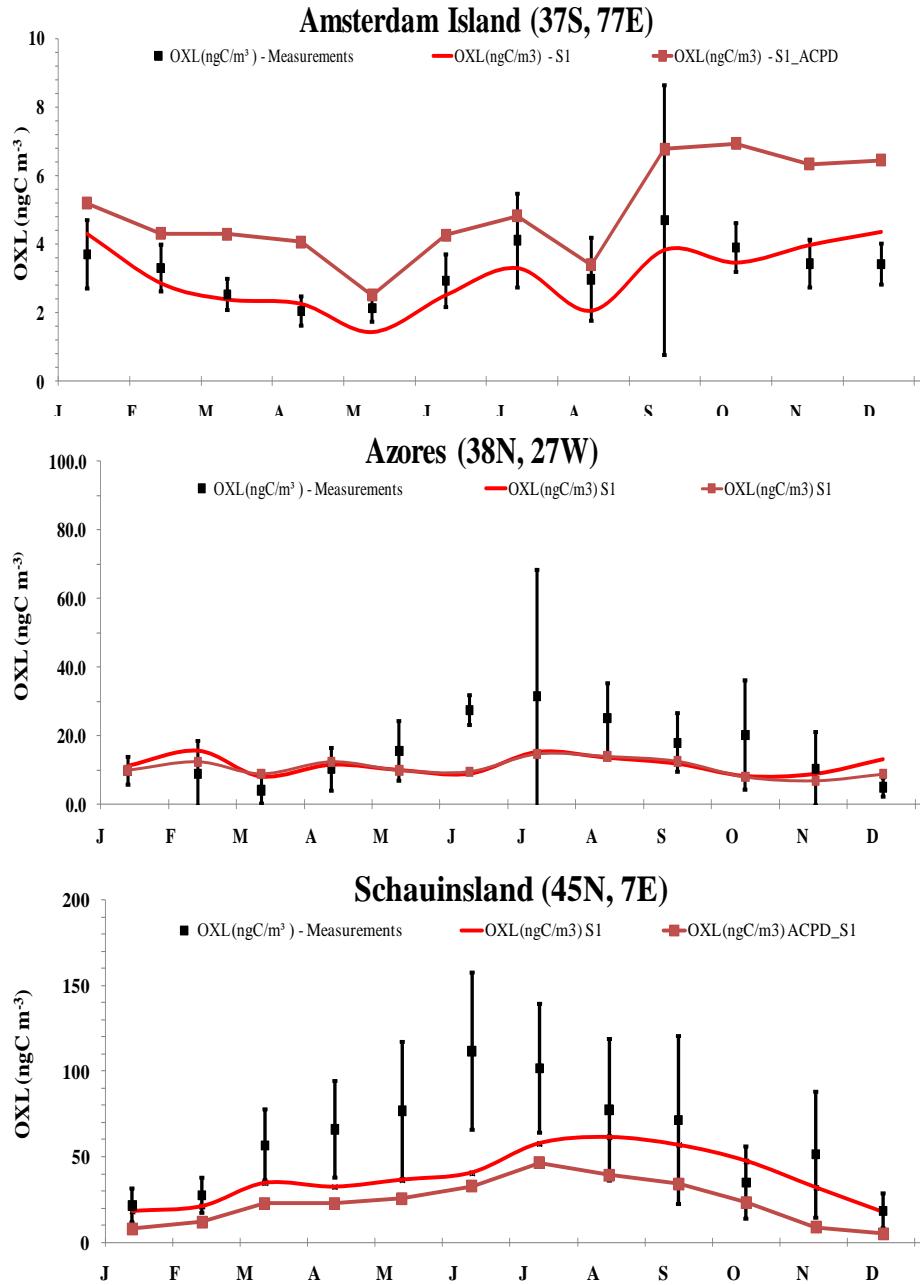
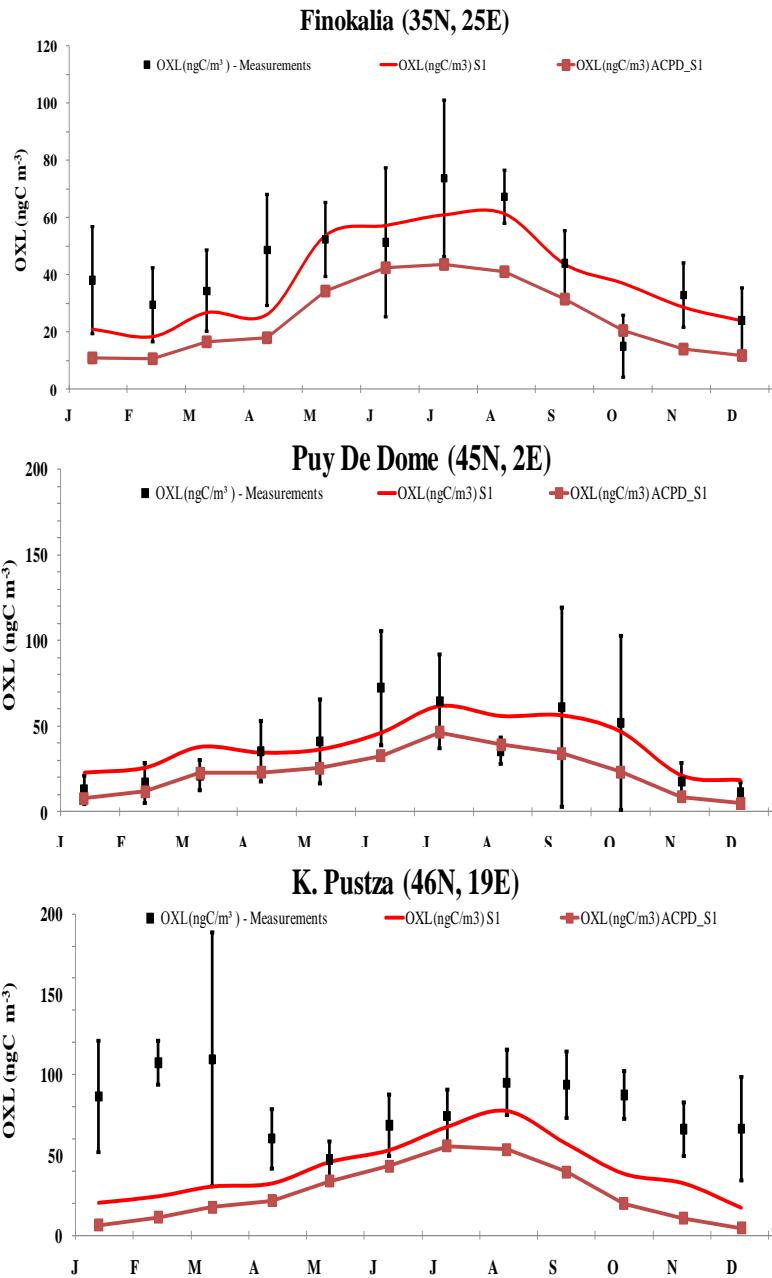
OXL Observation Sites



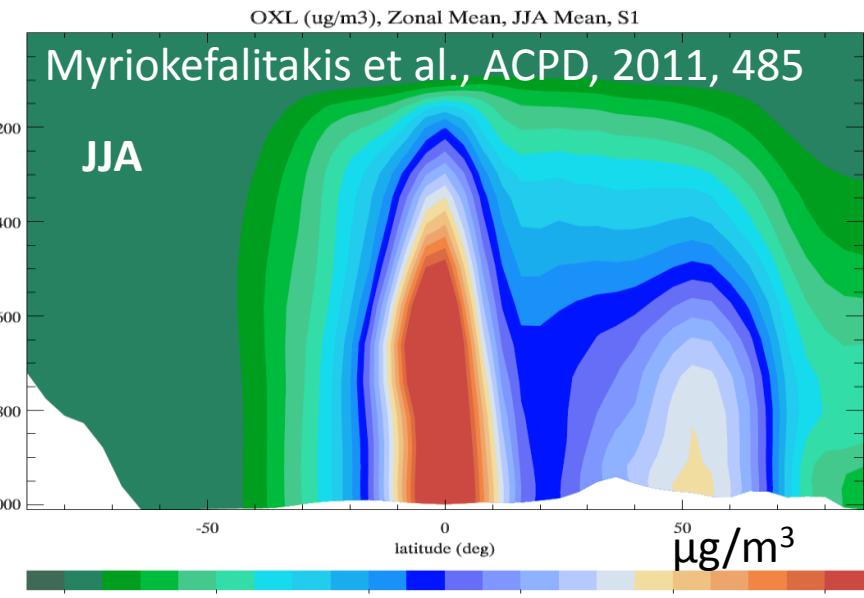
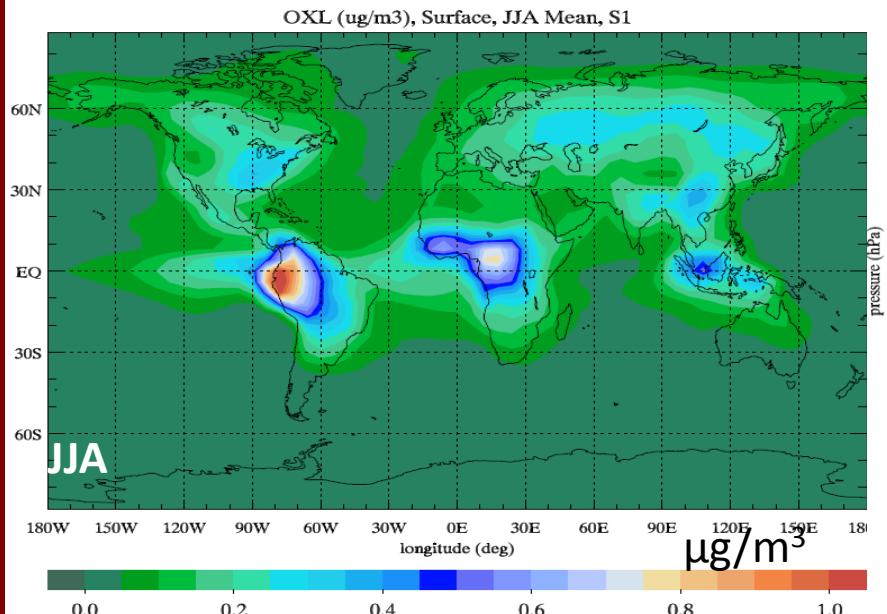
	Sites	Simulation	Slope	r ²	N	a	F-value
Global	Urban	S1	0.64 ± 1.1 9	-0.04	19	0.05	0.290
	Rural	S1	0.92 ± 0.2 4	0.18	64	0.05	14.842
	Marine	S1	1.13 ± 0.1 7	0.46	50	0.05	42.653
	Rural + Marine	S1	1.16 ± 0.1 4	0.36	114	0.05	65.598
	All Observations	S1	1.29 ± 0.2 0	0.23	133	0.05	41.346



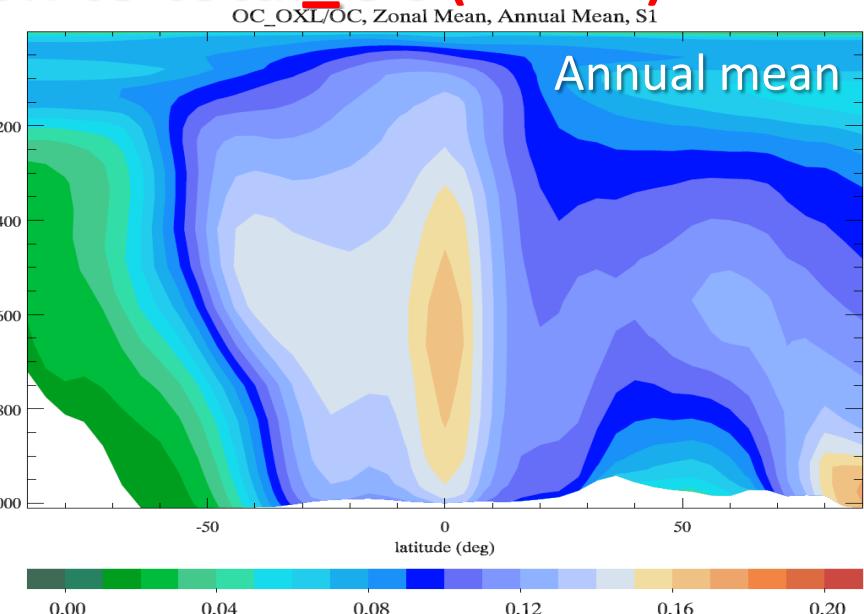
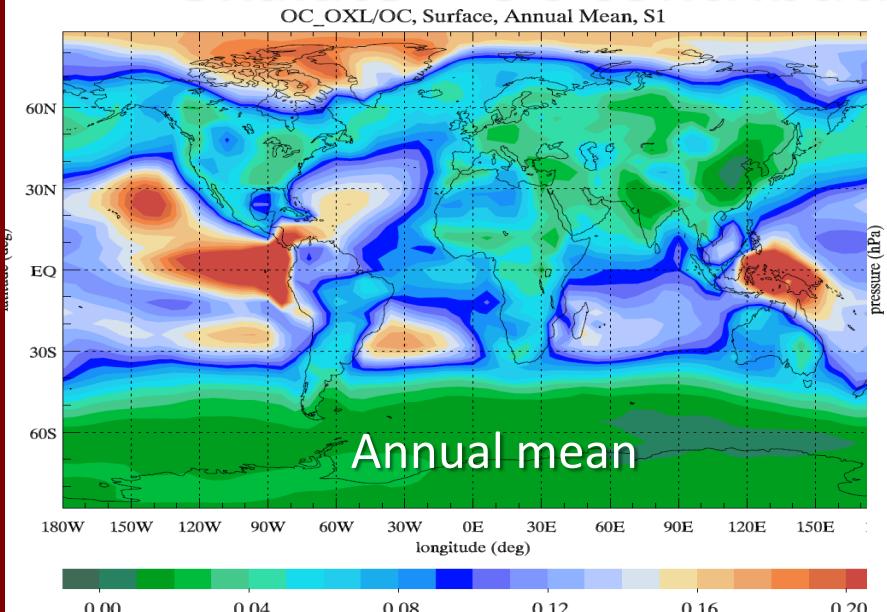
OXL Validations – OLD(ACPD) Vs. Revised Simulations



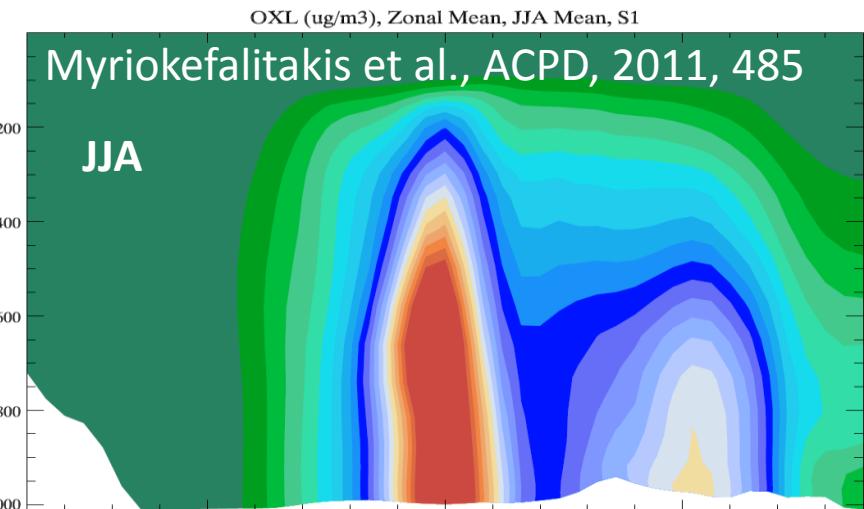
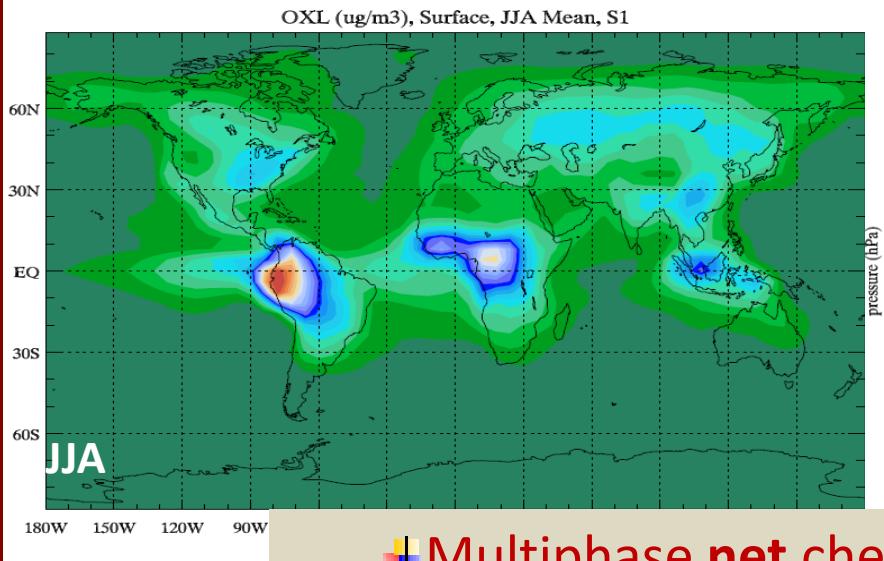
Global Oxalate Distributions



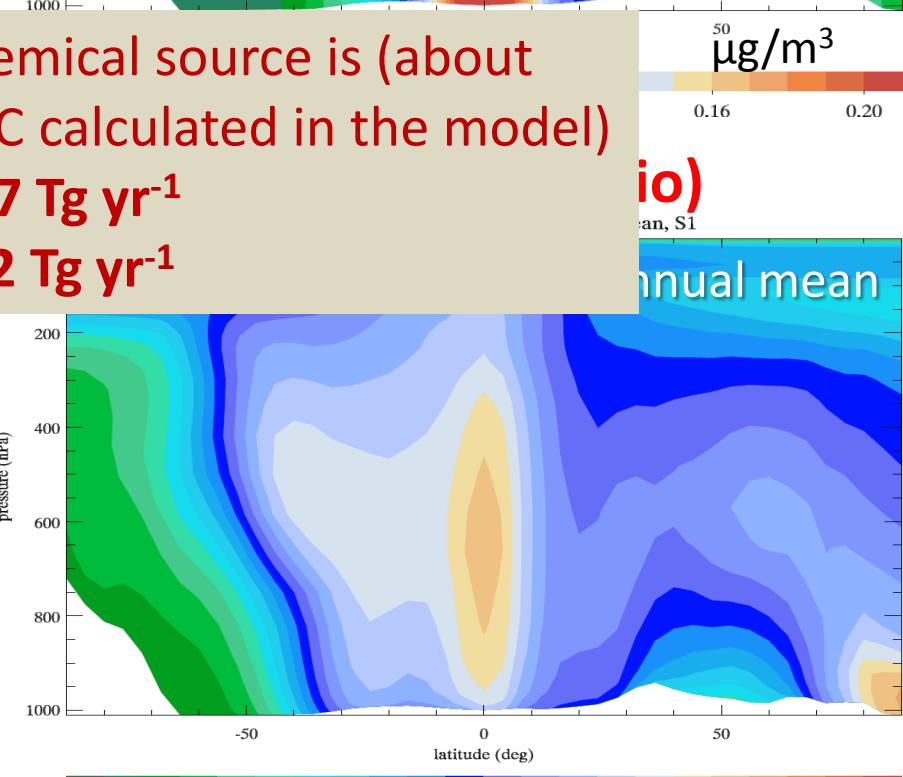
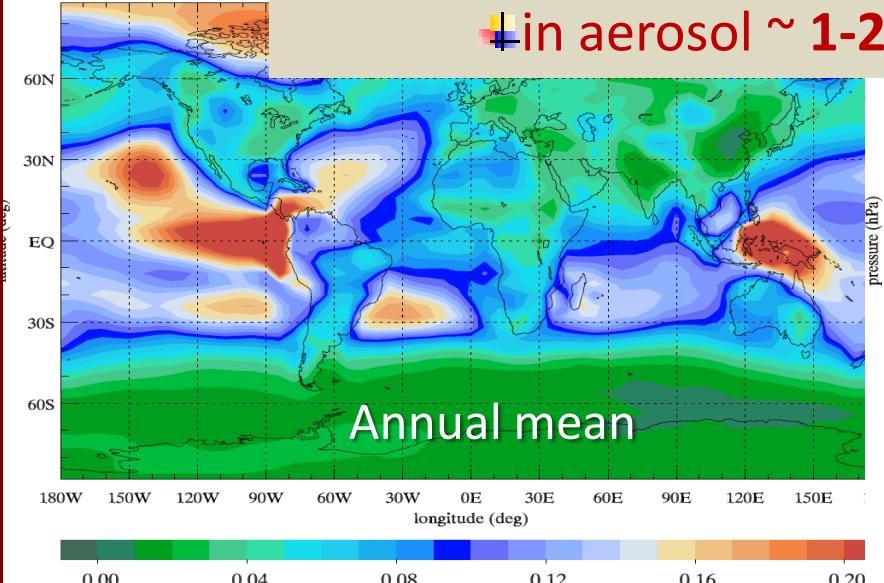
Oxalate – OC contribution to total_OC (C-ratio)



Global Oxalate Distributions



- Multiphase net chemical source is (about 5%-9% of total WSOC calculated in the model)
- In-cloud ~21-37 Tg yr⁻¹
- In aerosol ~ 1-2 Tg yr⁻¹



1. Multiphase chemistry (Stelios)
2. Interannual model evaluation & AEROCOM
(Nikos)
3. Other ongoing and future activities (Maria)

Changes in TM4ECPL

anthropogenic emissions updated to CIRCE database (2000 – 2010)

Interannual 2000 – 2005, 2010 (projection), 2006-2009 interpolated

0.1x0.1 ascii files → 1x1 hdf files

- land emissions (2d)

BC, CO, NH₃, NOx, OC, SO₂, NMVOC

- ship emissions (2d)

BC, CO, NOx, OC, SO₂, NMVOC

- aircraft emissions (3d)

BC, CO, NOx, OC, SO₂, NMVOC

All 3 categories with speciated NMVOC

speciation based on POET NMVOC speciation

species: acetone, butane, butene, C₂H₂, C₂H₅OH, C₂H₆, C₃H₆, C₃H₈, CH₂O, CH₃CHO, CH₃OH, MEK, toluene, benzene, xylene

Global Fire Emission Database v2 (1998 – 2008)

Gridded 1x1 netcdf files used, already speciated nmvocs (van der Werf)

Vertical elevation of Biomass Burning emissions based on Dentener et al, 2006

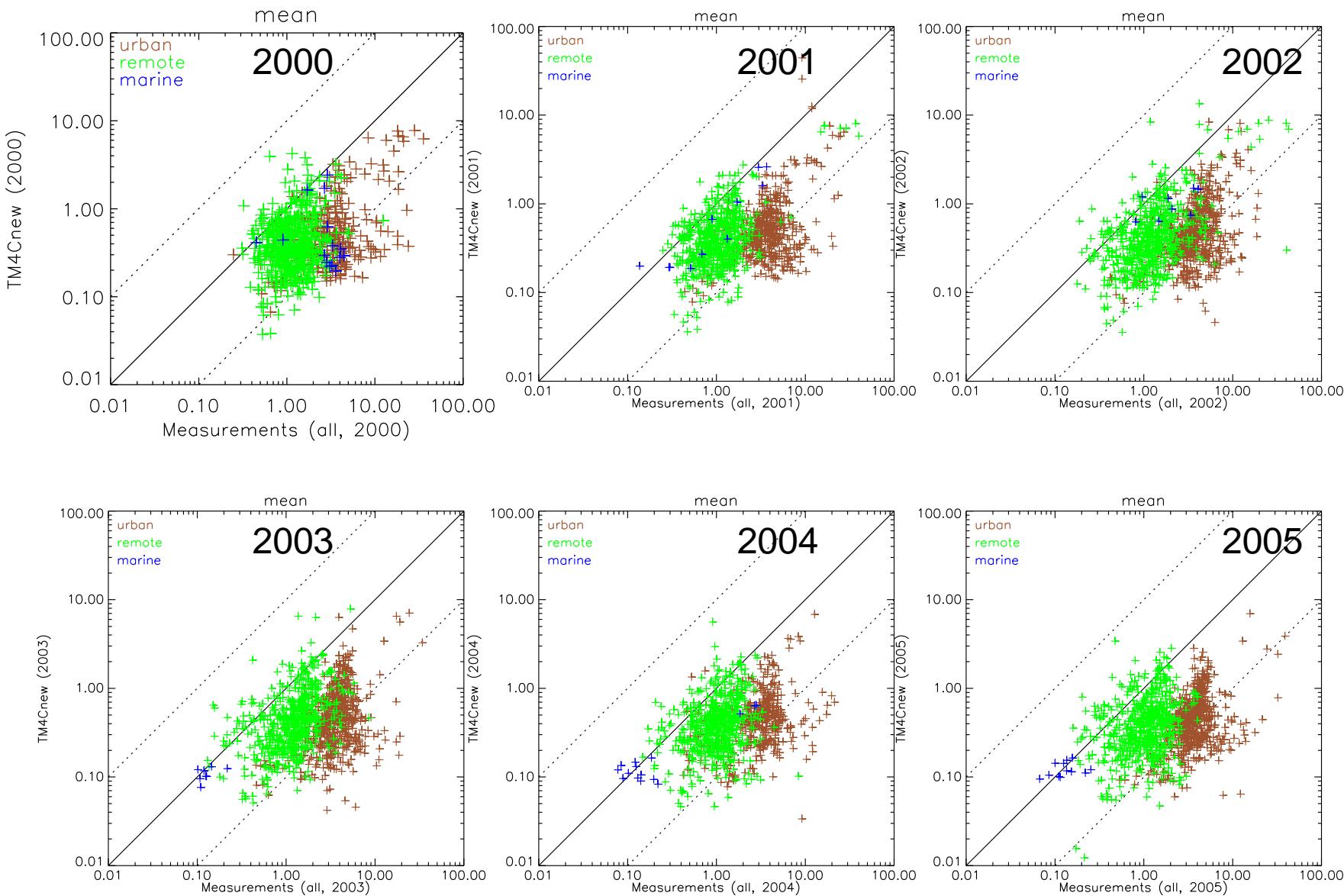
AEROCOM runs re-done

Interannual 6x4 (2000-2007)

2006 3x2

Model evaluation

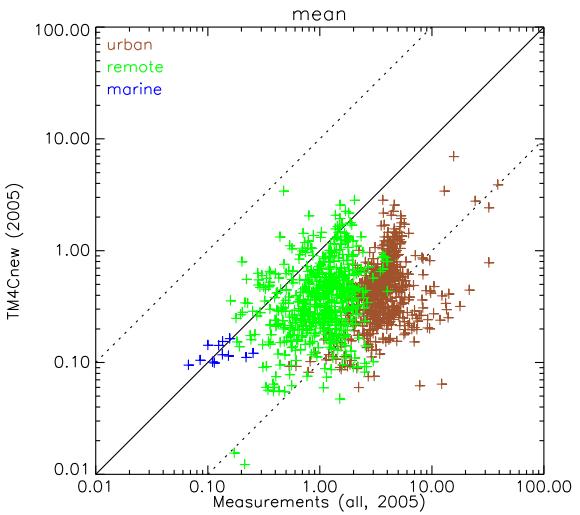
updated version (CIRCE anthropogenic, GFED v2, ERA meteorology,
Organic Carbon (comparison at all available stations) (ug/m3)



Organic Carbon -evaluation

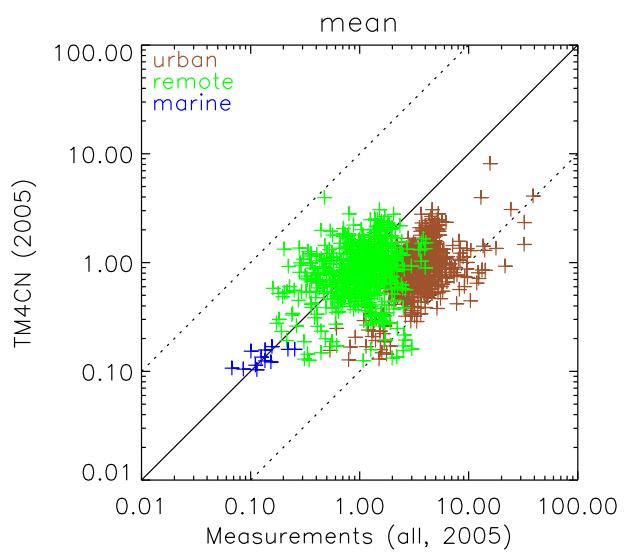
Model evaluation after the changes

- CIRCE anthropogenic
- Poet biogenic
- GFED v2
- operational meteo data



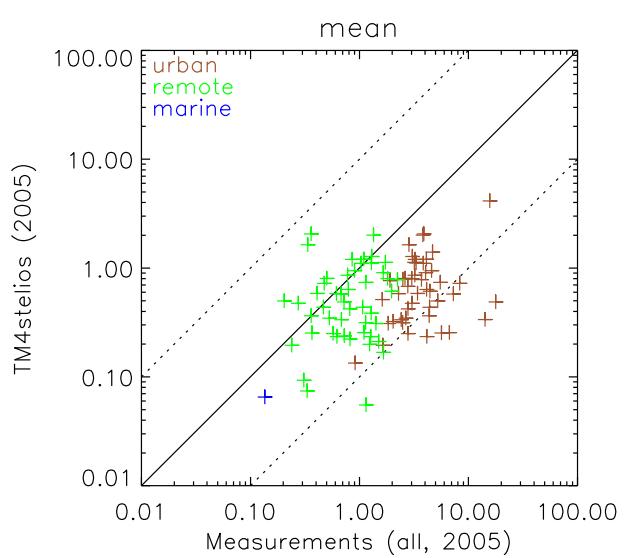
Model evaluation after the changes

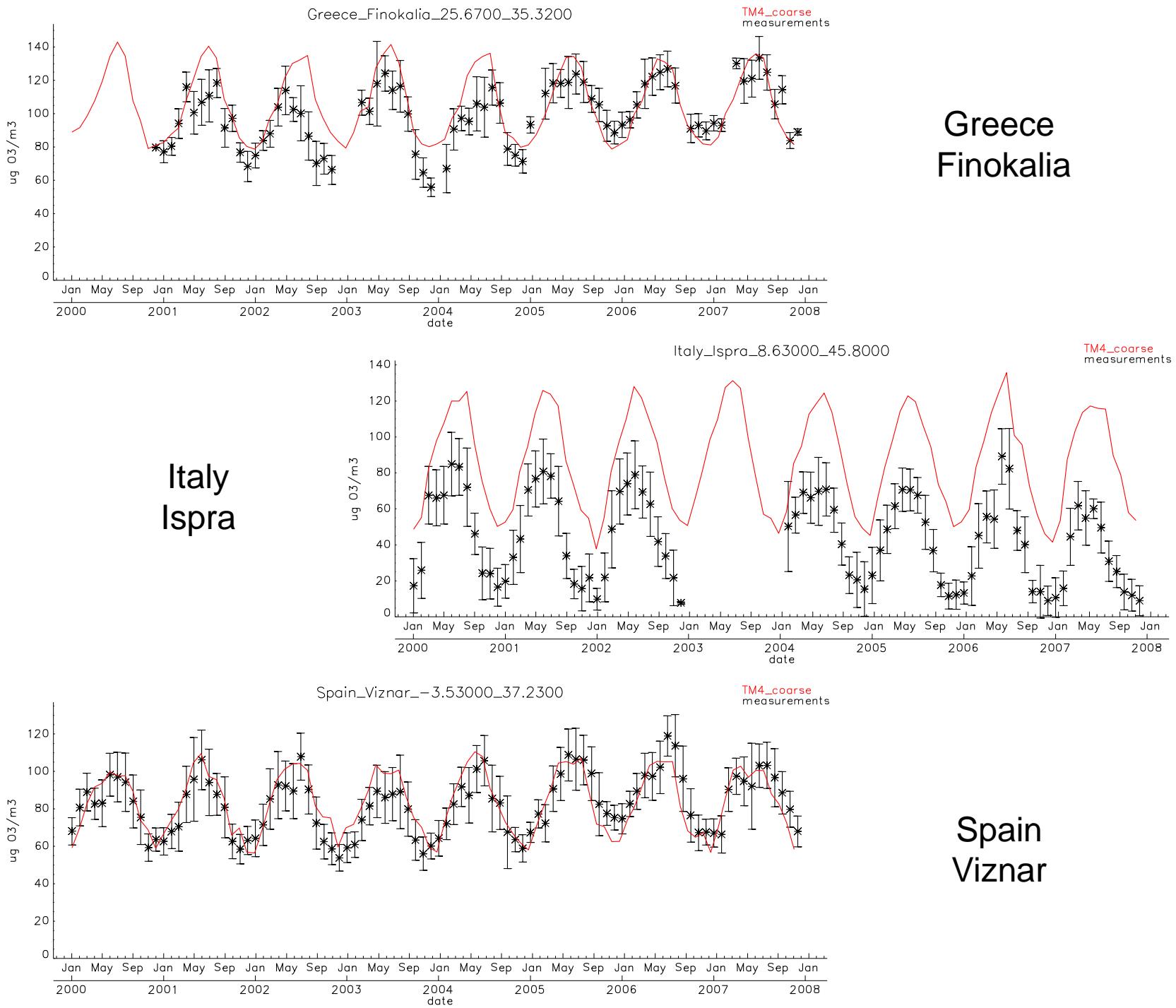
- CIRCE anthropogenic
- Poet biogenic
- GFED v2
- ERA meteo data



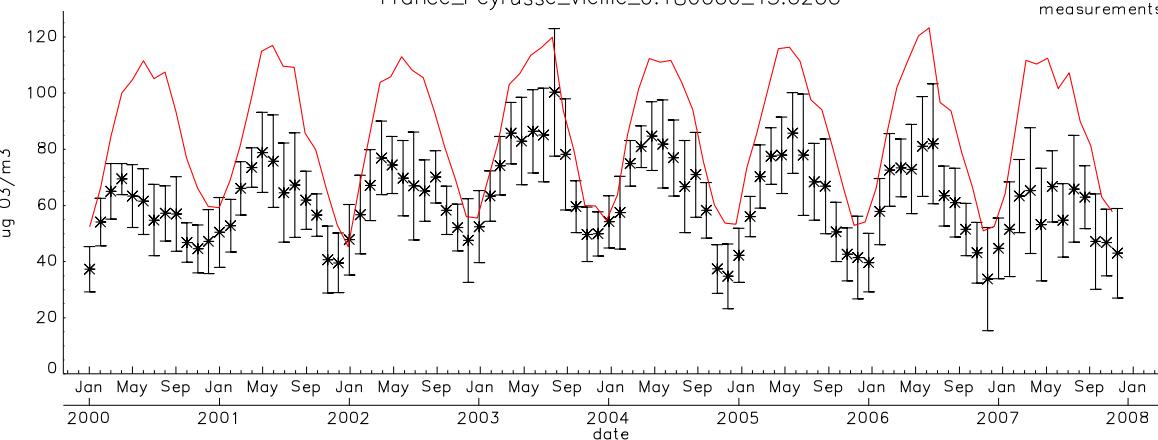
Model evaluation before the changes

- POET anthropogenic
- Poet biogenic
- GFED v2 CO and BC
- operational meteo data





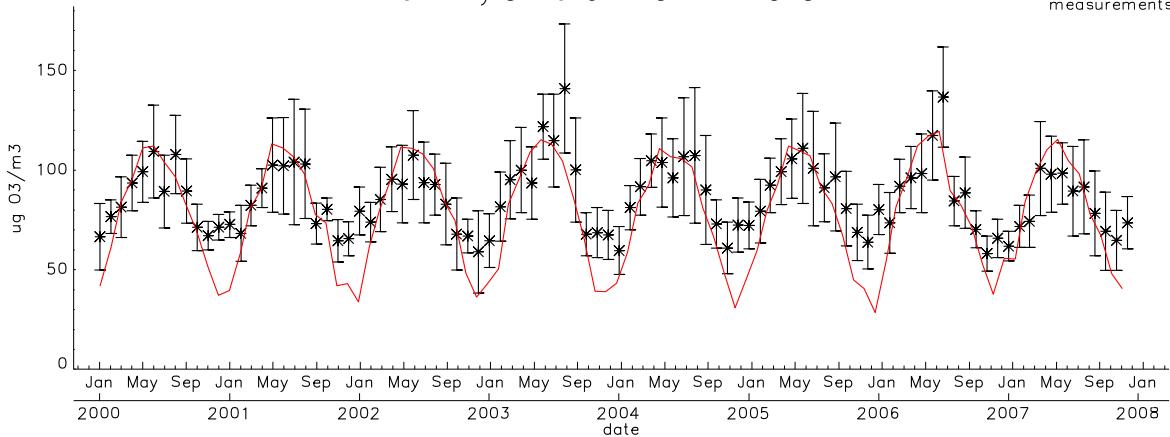
France Peyrusse Vieille



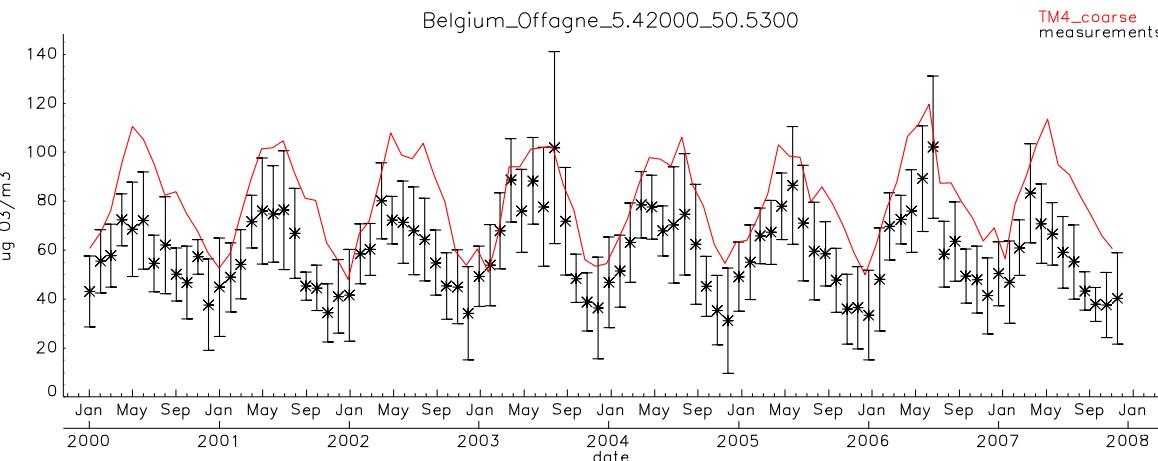
Germany_Schauinsland_8.42000_-48.7800

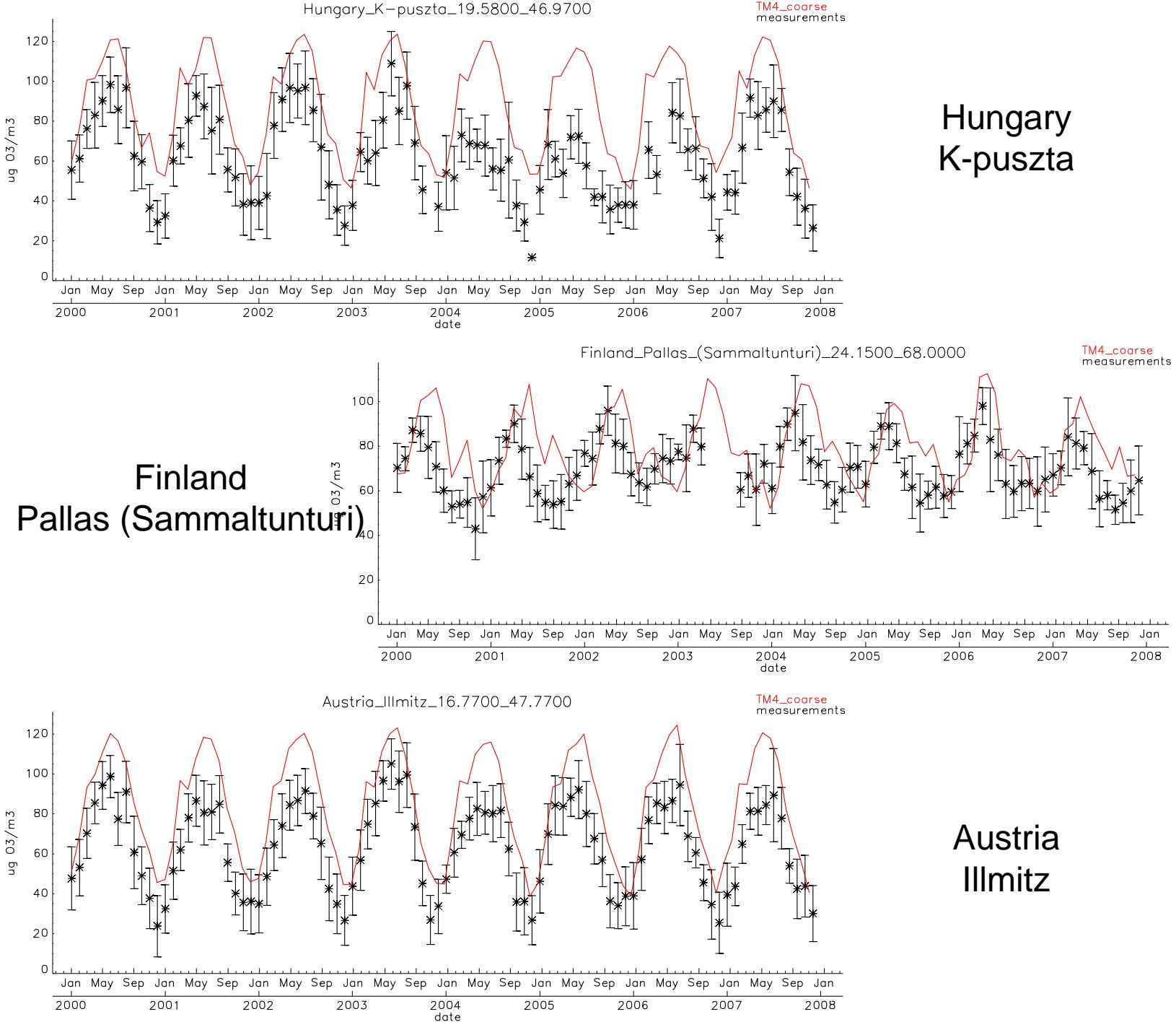
TM4_coarse
measurements

Germany Schauinsland

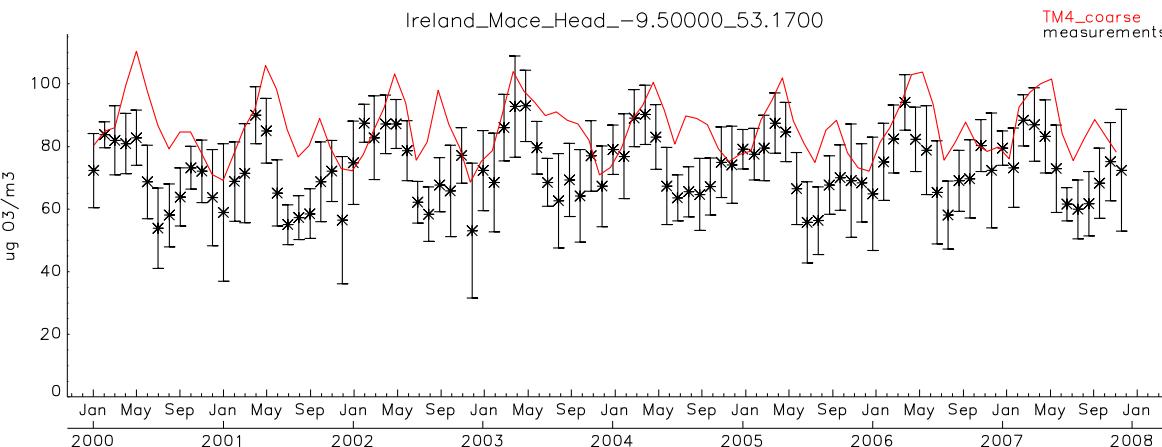


Belgium Offagne

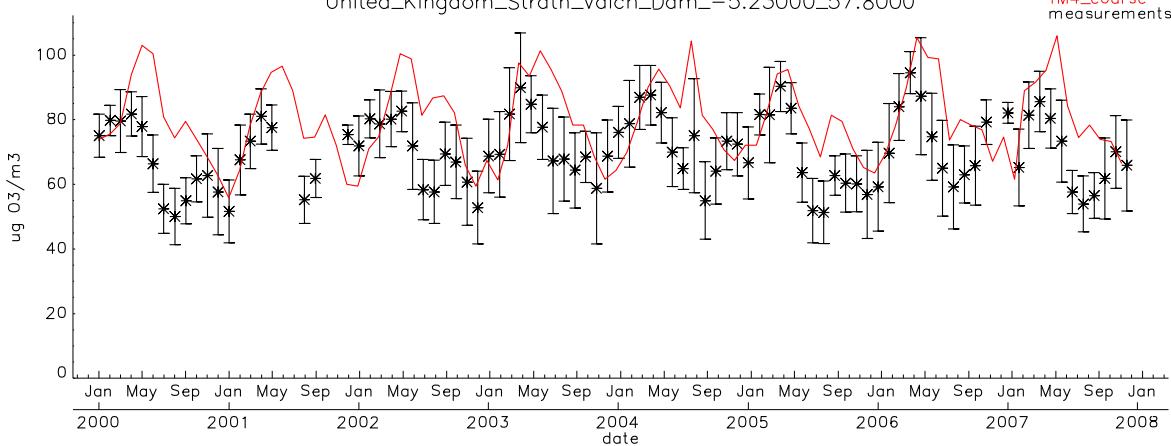




Ireland Mace Head

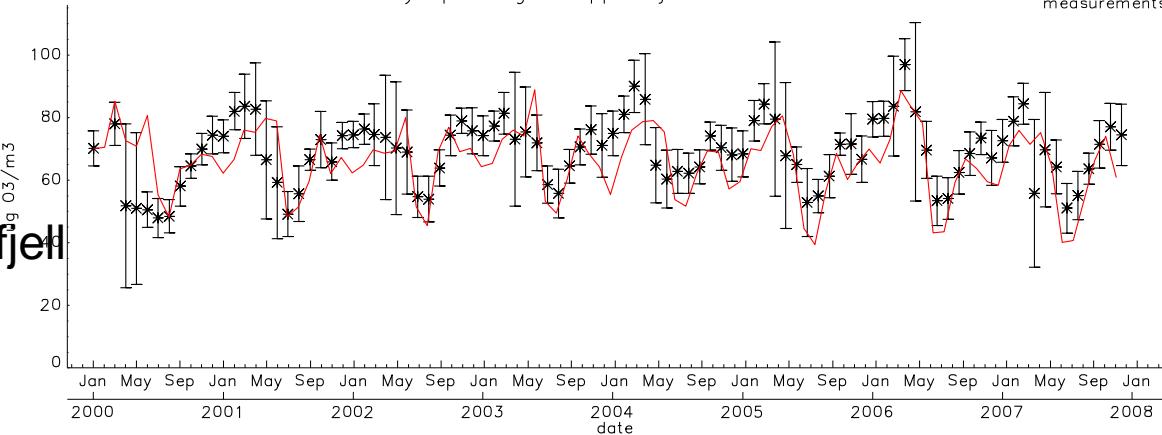


United_Kingdom_Strath_Vaich_Dam_-5.23000_57.8000



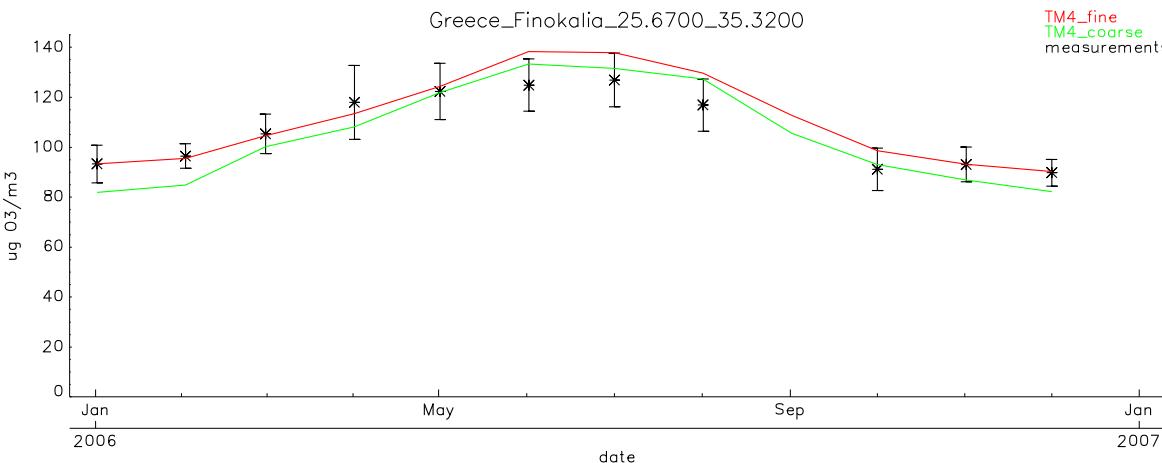
United Kingdom Strath Vaich Dam

Norway_Spitsbergen_Zeppelinfjell_11.8800_78.9000



Norway Spitsbergen Zeppelinfjell

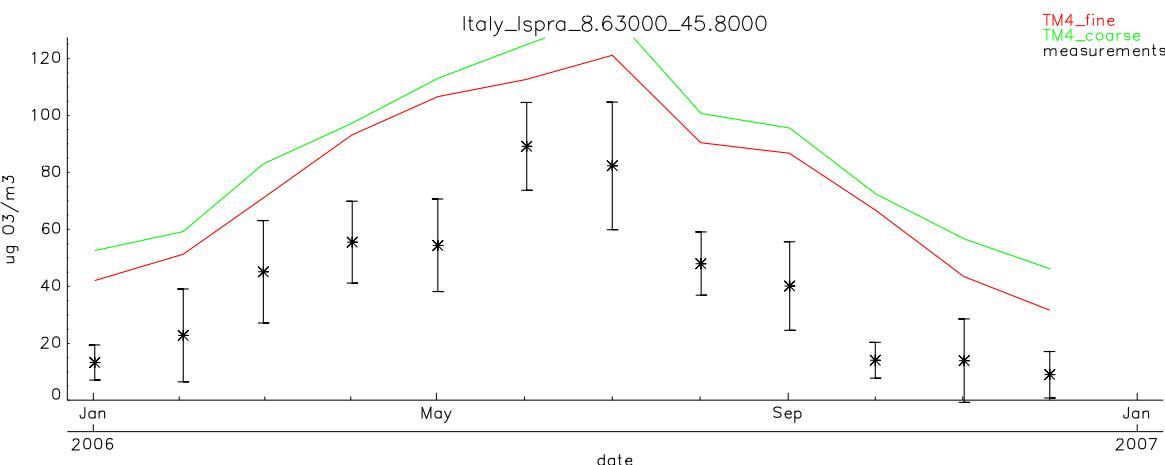
TM4_coarse
measurements



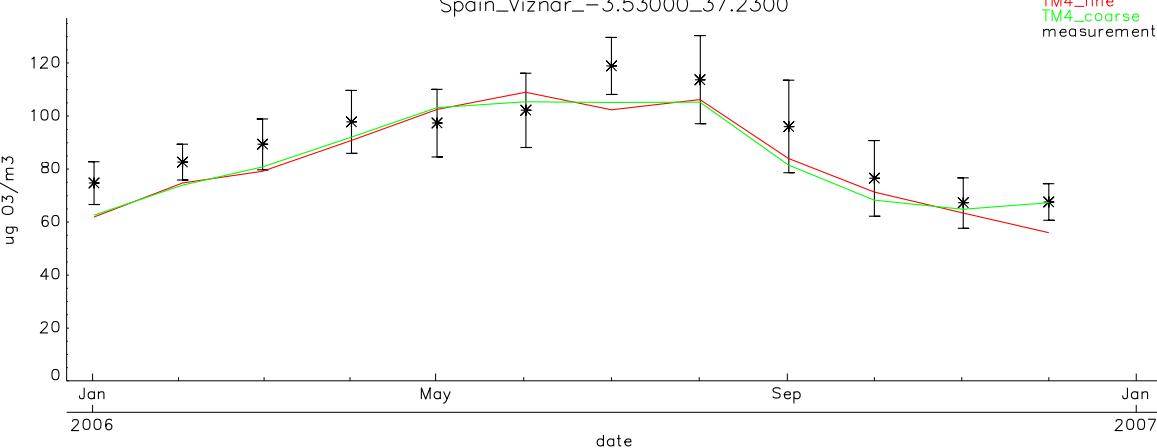
TM4-ECPL 3x2
TM4-ECPL-6x4
measurements

Greece
Finokalia

Italy
Ispra

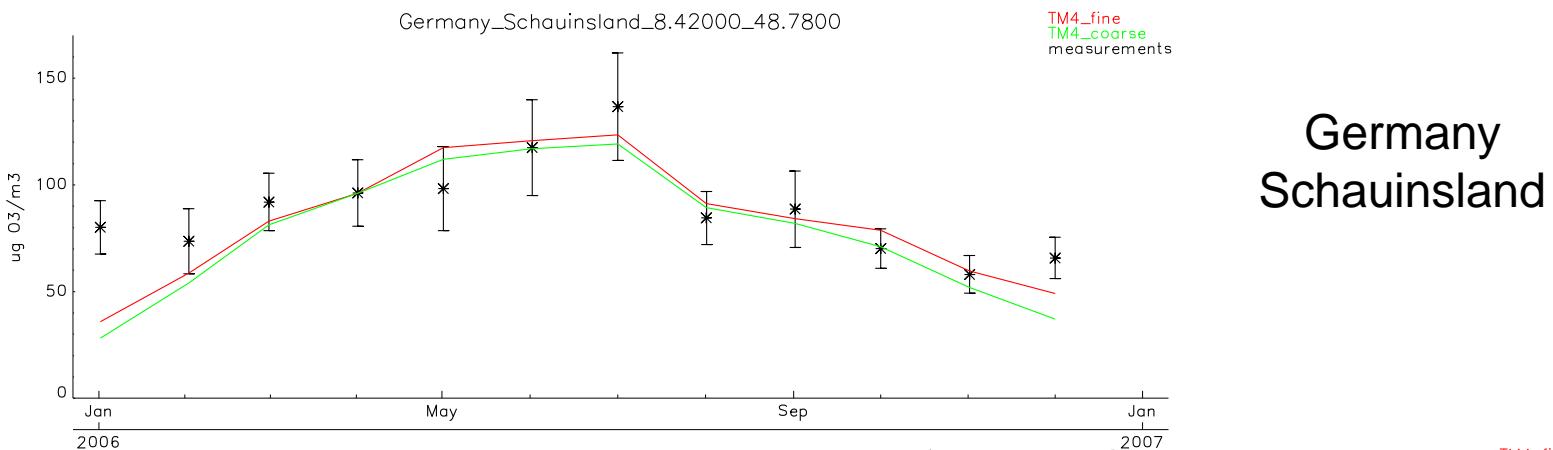
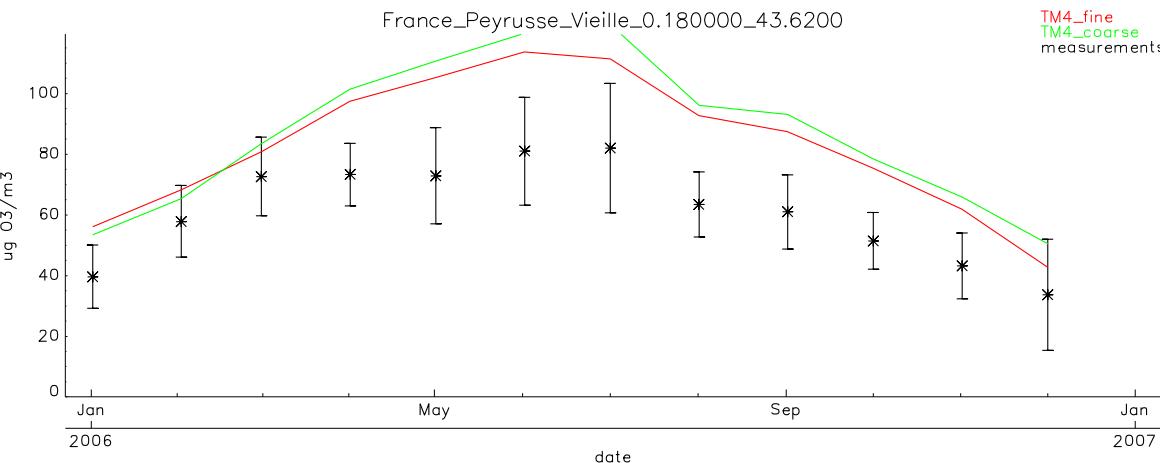


Spain
Viznar

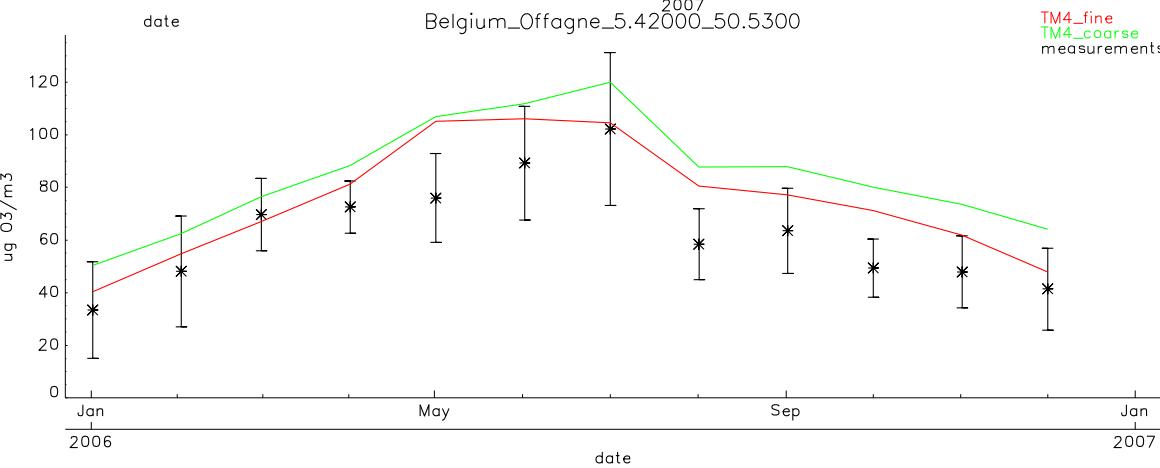


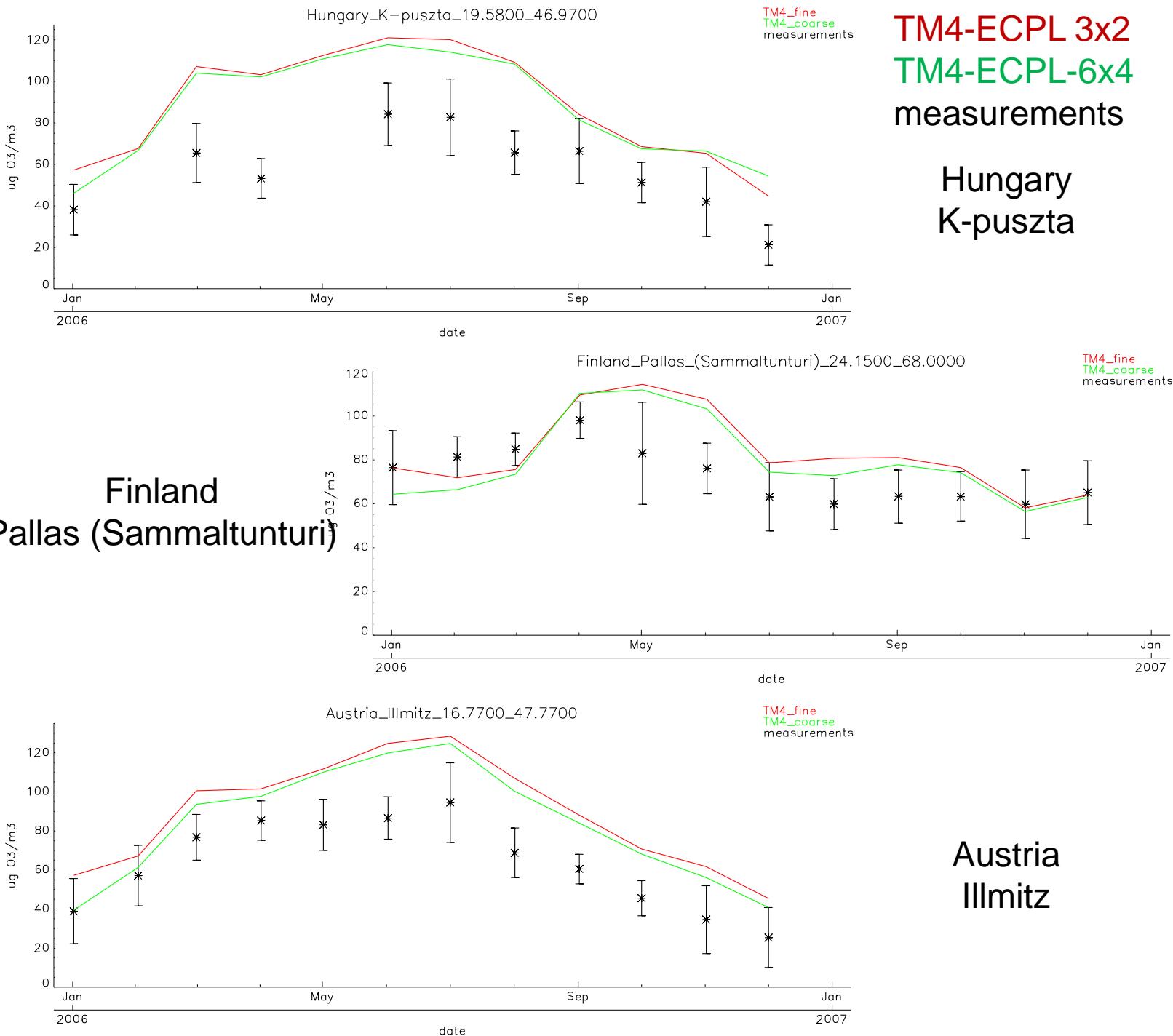
TM4-ECPL 3x2
TM4-ECPL-6x4
measurements

France Peyrusse Vieille



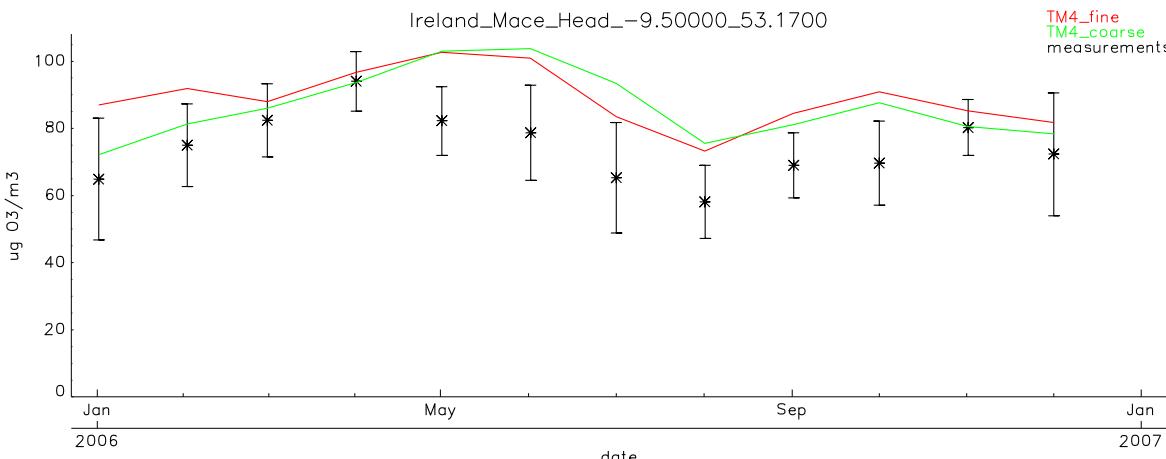
Belgium Offagne



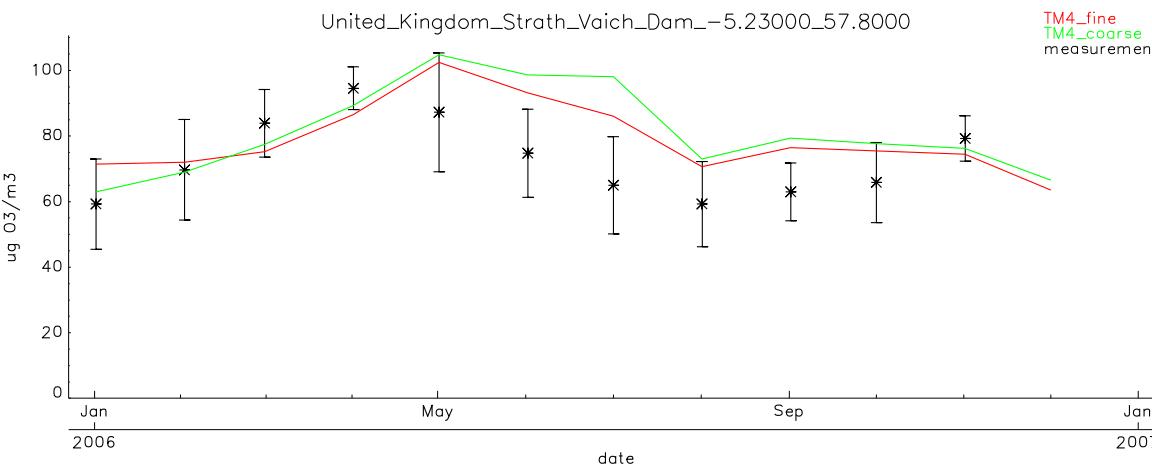


TM4-ECPL 3x2
TM4-ECPL-6x4
measurements

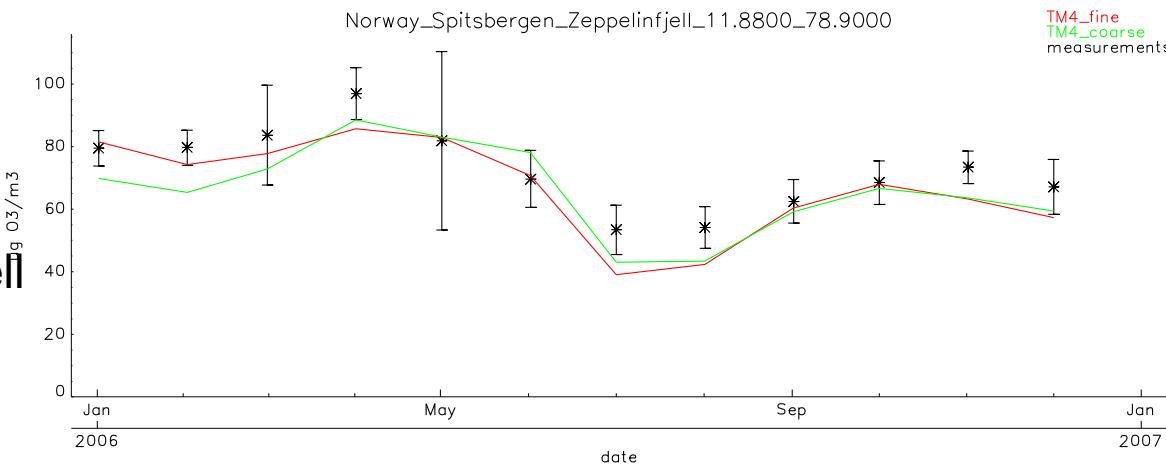
Ireland Mace Head



United Kingdom Strath Vaich Dam



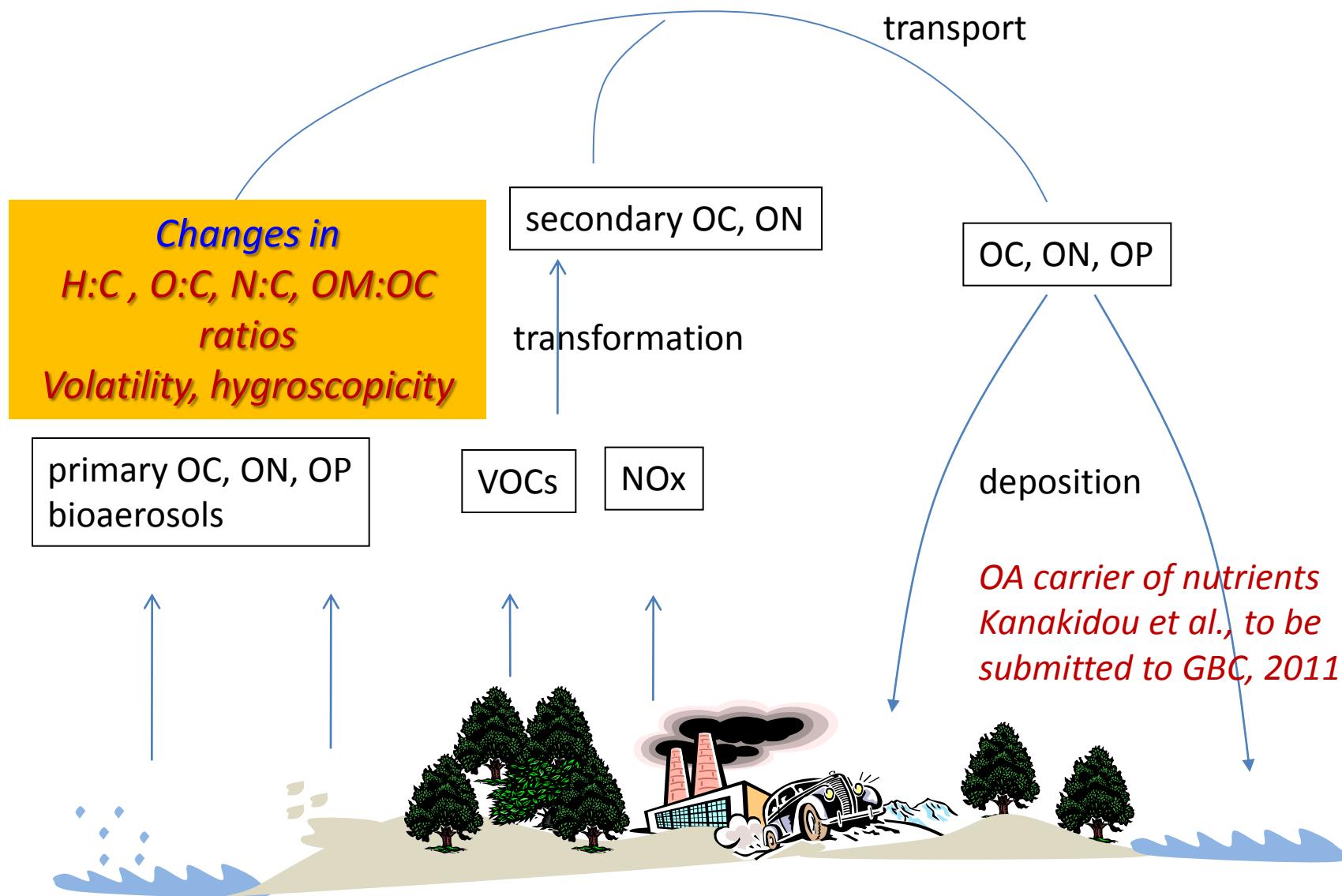
Norway Spitsbergen Zeppelinfjell



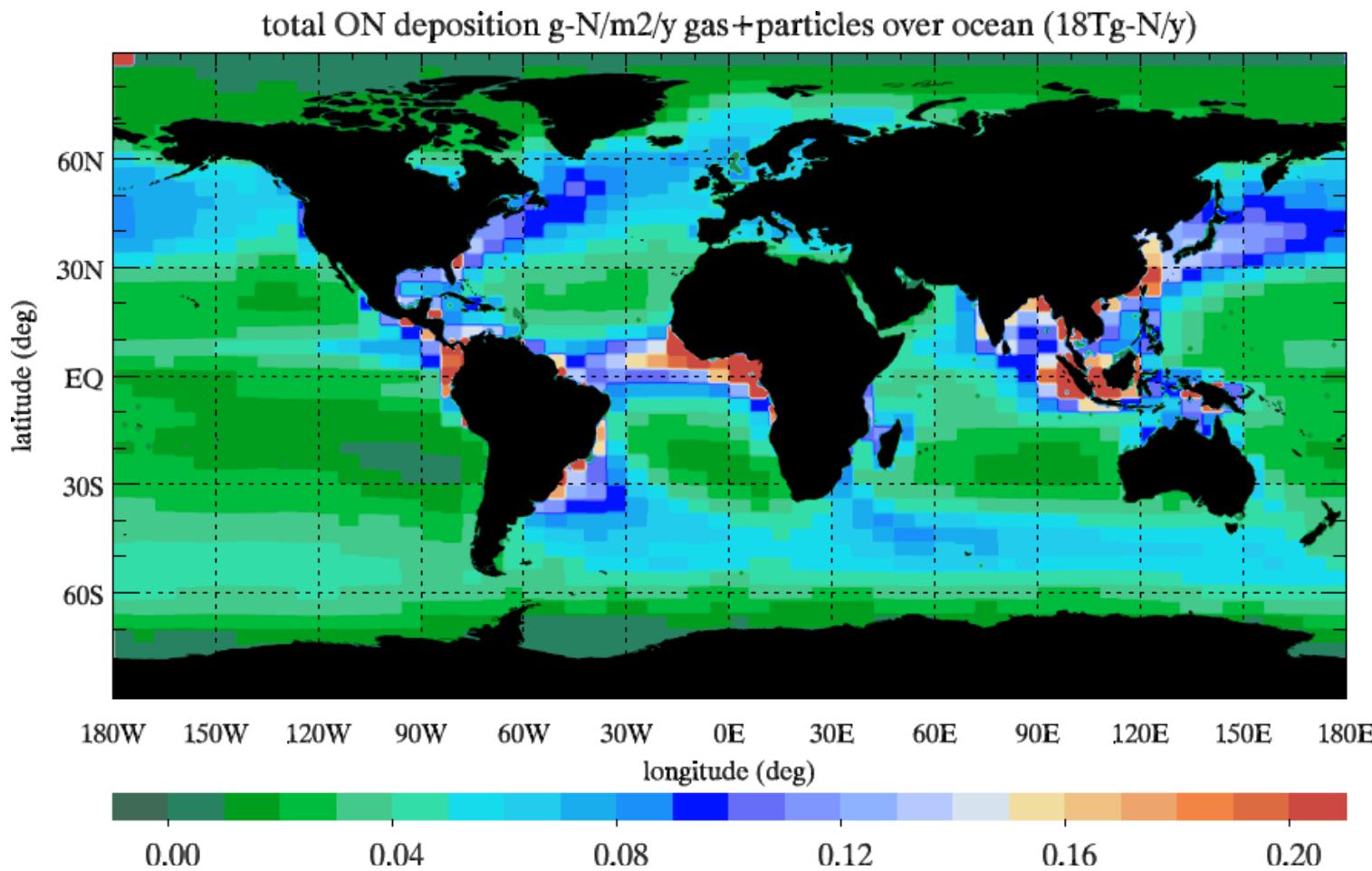
1. Multiphase chemistry (Stelios)
2. Interannual model evaluation & AEROCOM (Nikos)
3. Other ongoing and future activities (Maria)

Organic aerosol - new perspectives – based on element ratios

transformation



Organic aerosol - new perspectives – based on element ratios



OA carrier of nutrients

Kanakidou et al., to be submitted to GBC, 2011

Contribution to GESAMP WG38

Further work at ECPL

1. Sensitivity of model results to meteo, emissions, OA parameterisations
2. Interannual trends in oxidants + PM (past 20 years)
3. Oxidant and PM levels simulations – focus on Mediterranean
4. Budget analysis / import/export fluxes
5. Improve chemistry relevant parameterisations:
 1. OA parameterisations (volatility of POA, heterogeneous reactions, multiphase chemistry)
 2. oxidant chemistry (HOx recycling)
6. Sensitivity to emission location & height / geographic distribution/source receptor relationship/lifetime dependence
7. Impact of atmospheric deposition to the marine ecosystem (Mediterranean)