



# *The implementation of an explicit gas-phase chemical scheme in TM5-MP*

**Stelios Myriokefalitakis (UU), Maarten Krol (UU/WUR/SRON)  
Twan van Noije (KNMI), Philippe Le Sager (KNMI)**

**with the contribution of**

***Sander Houweling (VU/SRON), Maria Kanakidou (UoC) et al.***

## Part 1

1. Description of the new chemical scheme
2. Emissions
3. TM5MP - MOGUNTIA scheme coupling
4. TM5MP – KPP coupling

## Part 2

1. Ozone evaluation
2. CO evaluation
3. Conclusions and Further work...



# The MOGUNTIA chemical scheme in TM5-MP

➤ The MOGUNTIA chemical scheme (*Poisson et al., 2000; Myriokefalitakis et al., 2008, 2011*) is a rather explicit oxidation mechanism of Volatile Organic Compounds (VOCs) including biogenic and aromatic hydrocarbon chemistry:

➤ C<sub>1</sub>-C<sub>3</sub>: detailed oxidation

➤ *light alkanes (CH<sub>4</sub>, C<sub>2</sub>H<sub>6</sub> and C<sub>3</sub>H<sub>8</sub>, C<sub>4</sub>H<sub>10</sub>), light alkenes (C<sub>2</sub>H<sub>4</sub> and C<sub>3</sub>H<sub>6</sub>)*

➤ *CH<sub>3</sub>CHO, CHOCHO, HOCHCHO, CH<sub>3</sub>COCHO and CH<sub>3</sub>COCH<sub>3</sub>*

➤ *Methacrolein and Methylvinyl ketone*

➤ *NO<sub>3</sub> oxidation of aldehydes, alcohols, n-C<sub>4</sub>H<sub>10</sub>, and unsaturated hydrocarbons are also considered.*

➤ C<sub>(n≥4)</sub> (higher VOCs): based on n-butane oxidation

➤ Isoprene: detailed oxidation

➤ Terpenes: simplified chemistry based on isoprene oxidation

➤ Aromatics: simplified chemistry based on C<sub>(n≥4)</sub> oxidation

➤ Sulphur and ammonia chemistry as well as heterogeneous reactions as in the standard CB05 scheme



# VOC EMISSIONS in TM5-MP

Species	Long name	Emissions (Tg yr <sup>-1</sup> )
CO	Carbon Monoxide	1095.0
CH <sub>4</sub>	Methane	613.3
C <sub>5</sub> H <sub>8</sub> (ISOP)	Isoprene	572.5
CH <sub>3</sub> OH	Methanol	145.0
CH <sub>3</sub> CHO	Acetaldehyde	132.1
C <sub>10</sub> H <sub>16</sub> (TERP)	Terpenes	103.4
NH <sub>3</sub>	Ammonia	95.7
NO <sub>x</sub>	Nitrogen Oxides	70.9
C <sub>4</sub> H <sub>10</sub>	Butane and higher alkanes	59.9
CH <sub>3</sub> COCH <sub>3</sub> (ACET)	Acetone (includes all ketones except MEK from BB)	53.4
CH <sub>3</sub> SCH <sub>3</sub> (DMS)	Dimethylsulphide (DMS)	41.2
C <sub>3</sub> H <sub>6</sub>	Propene	37.2
C <sub>2</sub> H <sub>4</sub> (ETH)	Ethene	28.9
CH <sub>3</sub> COOH (MCOOH)	Acetic acid	26.3
CH <sub>3</sub> CH <sub>2</sub> OH (ETHOH)	Ethanol	26.0
C <sub>7</sub> H <sub>8</sub> (AROM)	Aromatics (lumped on toluene)	18.6
CH <sub>2</sub> O	Formaldehyde	16.6
C <sub>2</sub> H <sub>6</sub>	Ethane	12.1
HCOOH	Formic acid	10.9
C <sub>3</sub> H <sub>8</sub>	Propane	9.7
CHOCHO (GLY)	Glyoxal	8.5
HOCH <sub>2</sub> CHO (GLYAL)	Glycol-aldehyde	7.6
CH <sub>3</sub> CH <sub>2</sub> CHO (MGLY)	Methylglyoxal	5.1
C <sub>2</sub> H <sub>2</sub>	Acetylene	5.1
CH <sub>3</sub> CH <sub>2</sub> COCH <sub>3</sub> (MEK)	Methyl-ethyl-ketone (MEK)	3.3





# Coupling KPP with TM5-MP

- The Kinetic PreProcessor (KPP) is a software that automatically generates code that solves a chemical scheme (defined by input files).
- KPP software tools can be applied to any kinetic mechanism and provides a comprehensive suite of stiff numerical integrators (stiff solvers).
- The KPP framework allows to easily incorporate various solvers (e.g. **radau5**, **Isodes**, **runge\_kutta**, **rosenbrock**) with the recommended solver to be the Rosenbrock Rodas3 (*Sandu and Sander, 2006*).
- KPP has been successfully integrated with major models (e.g., CMAQ, GEOS-Chem, STEM, ECHAM5/MESSy, WRF-Chem, C-IFS...) and provides a good combination of accuracy and efficiency.

## KPP-generated code...

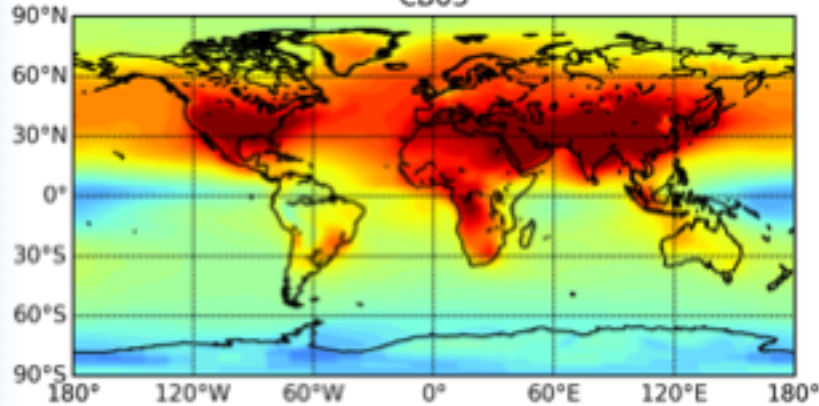
- 1) Write the gas- (and aqueous) phase mechanisms in KPP format
- 2) Create a new driver to interface the produced KPP files with the TM5 code
- 1) Preliminary tests show that the KPP-CB05 is **~14%** slower than the standard EBI-CB05 solver for a glb600x400-tropo25 configuration (00:23:52 against 00:21:06 (hh:mm:ss) for JAN 2015; nx:4, ny: 6).



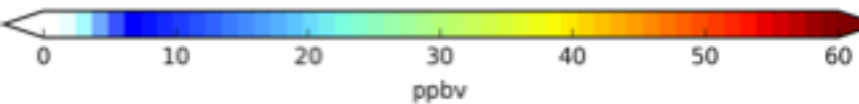
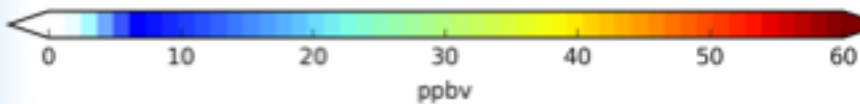
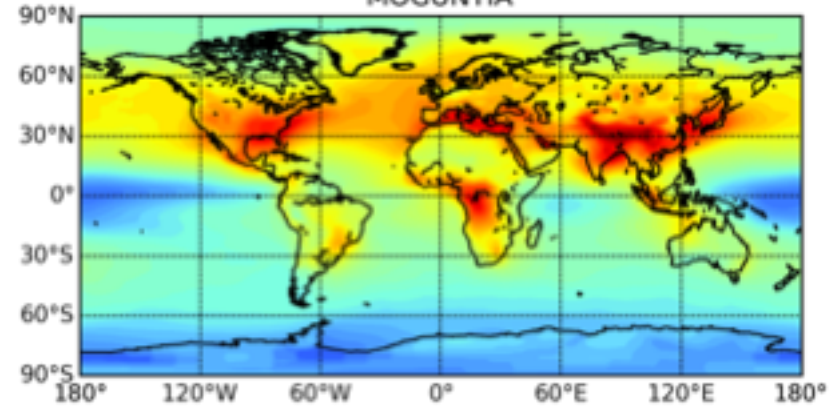


# Ozone: CB05 vs MOGUNTIA

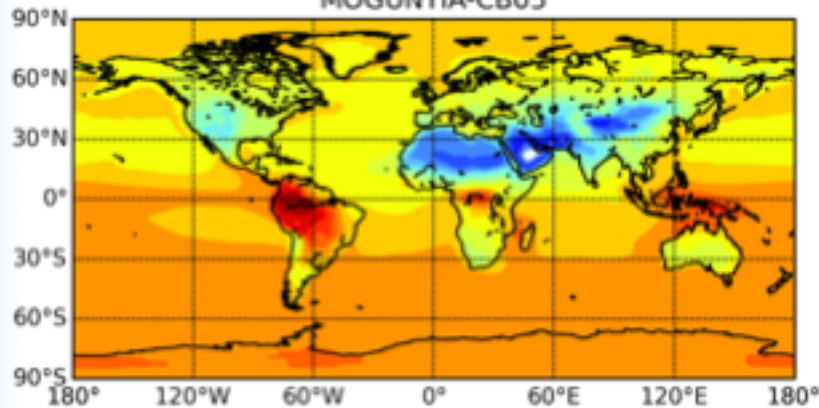
O3, Surface, Annual Mean  
CB05



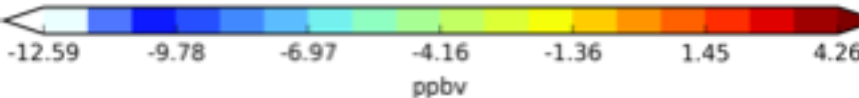
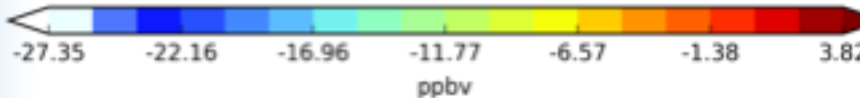
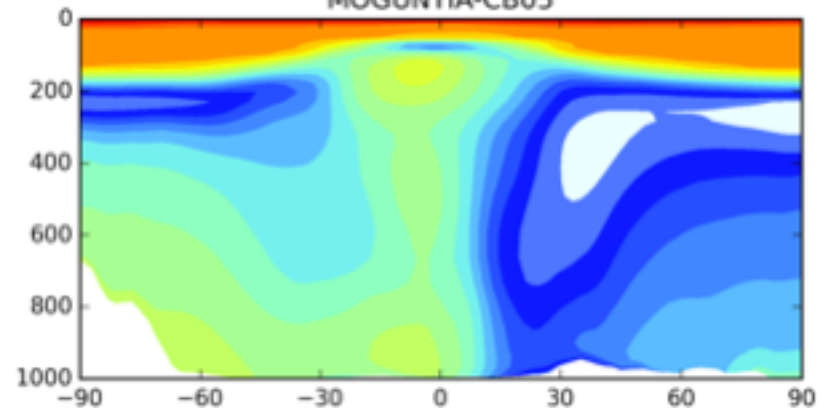
O3, Surface, Annual Mean  
MOGUNTIA



O3, Surface, Annual Mean  
MOGUNTIA-CB05

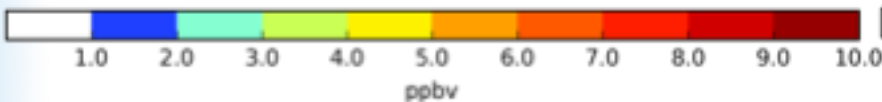
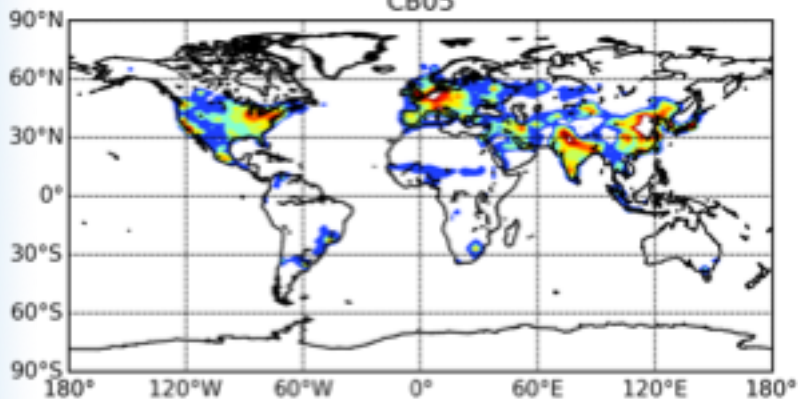


O3, Zonal Mean, Annual Mean  
MOGUNTIA-CB05

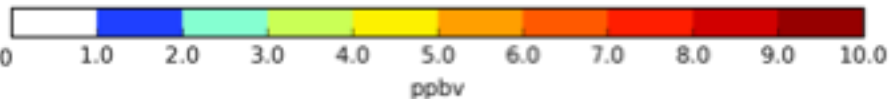
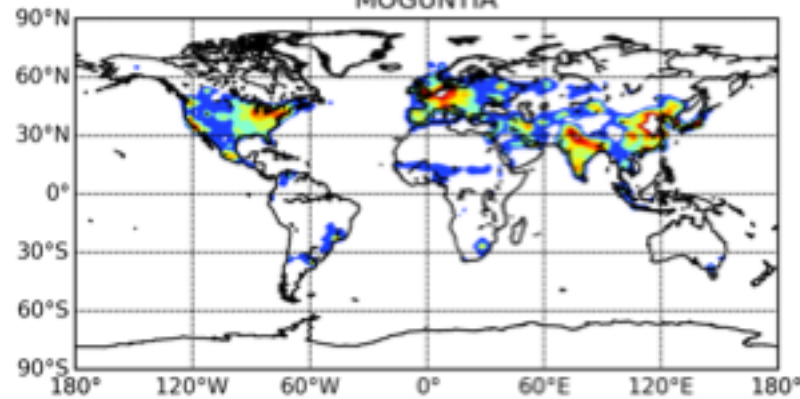


# NOx: CB05 vs MOGUNTIA

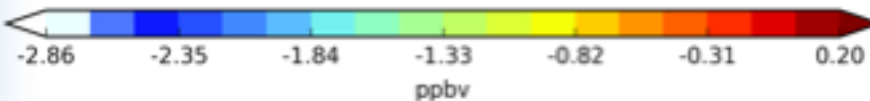
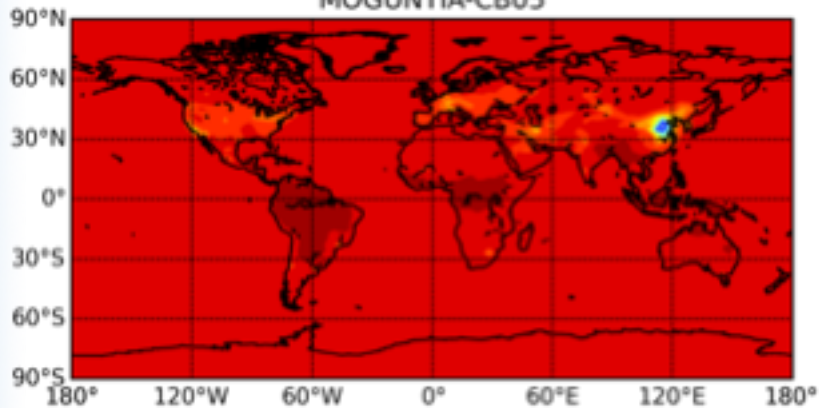
NOx, Surface, Annual Mean  
CB05



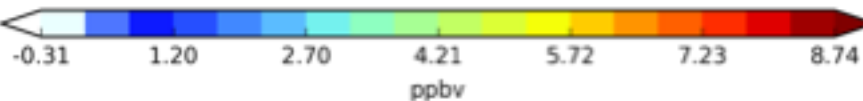
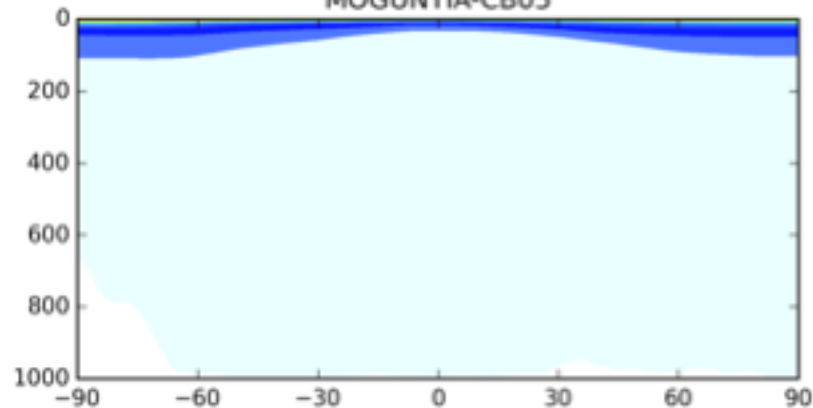
NOx, Surface, Annual Mean  
MOGUNTIA



NOx, Surface, Annual Mean  
MOGUNTIA-CB05



NOx, Zonal Mean, Annual Mean  
MOGUNTIA-CB05

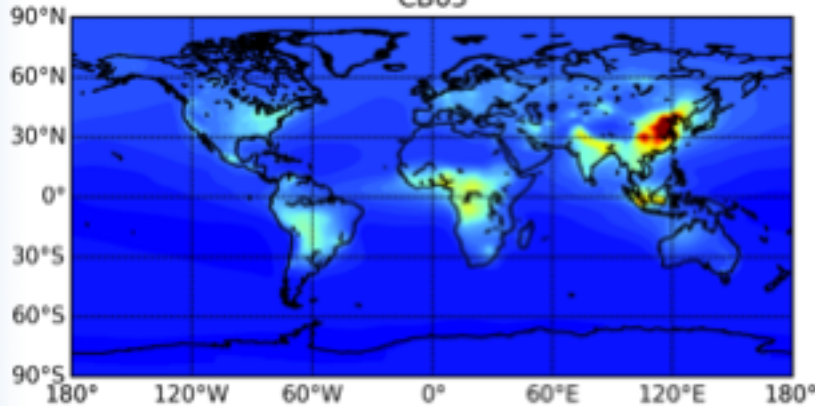




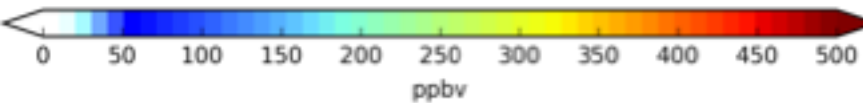
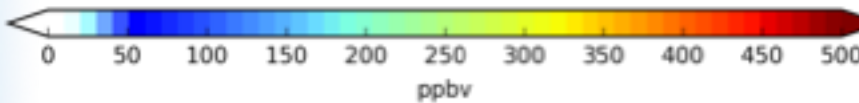
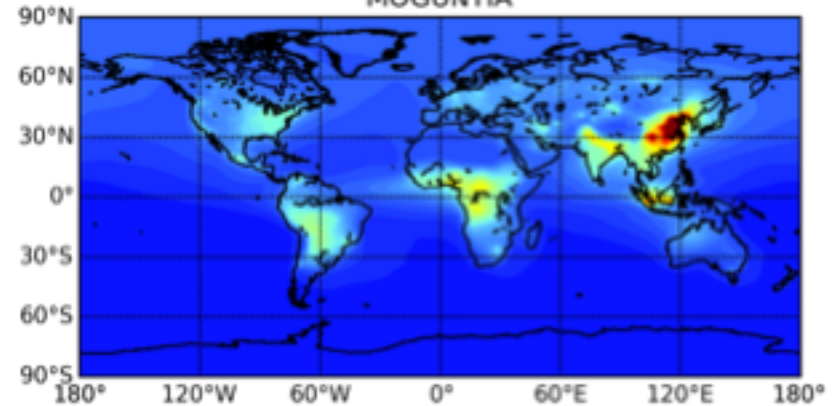
# CO: CB05 vs MOGUNTIA

Global Lifetime\*  
64 days → 67 days

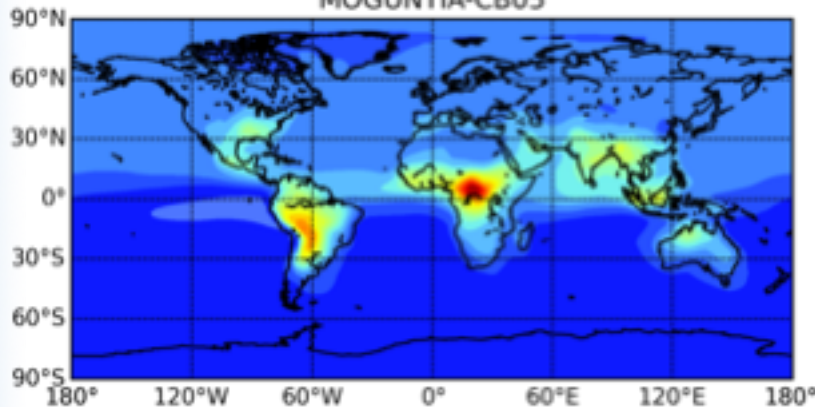
CO, Surface, Annual Mean  
CB05



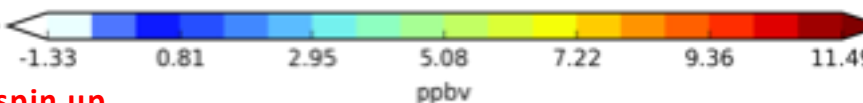
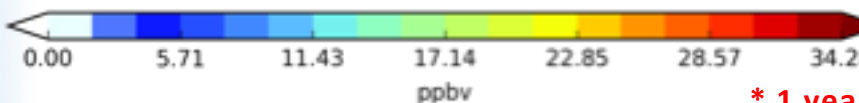
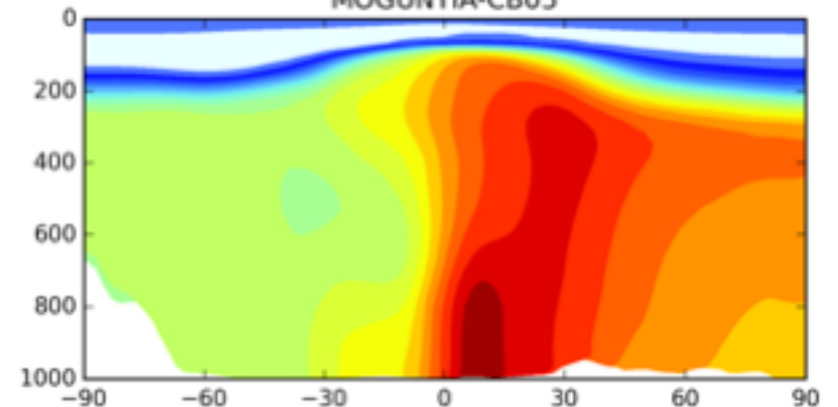
CO, Surface, Annual Mean  
MOGUNTIA



CO, Surface, Annual Mean  
MOGUNTIA-CB05



CO, Zonal Mean, Annual Mean  
MOGUNTIA-CB05

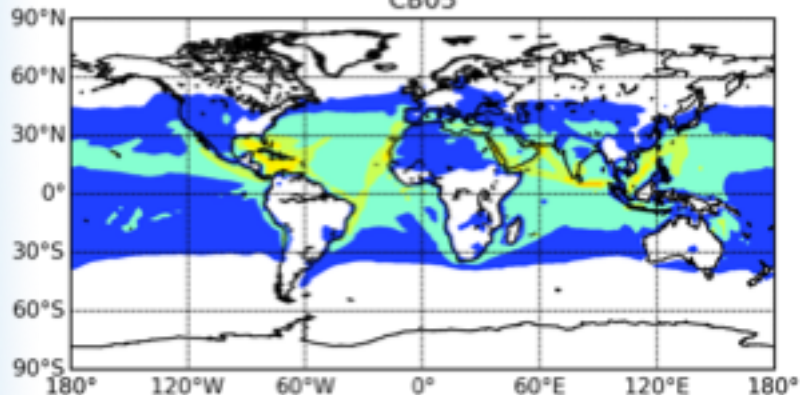


\* 1 year spin up

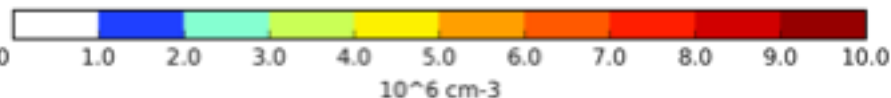
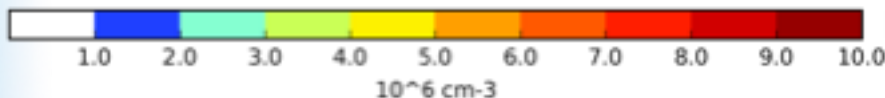
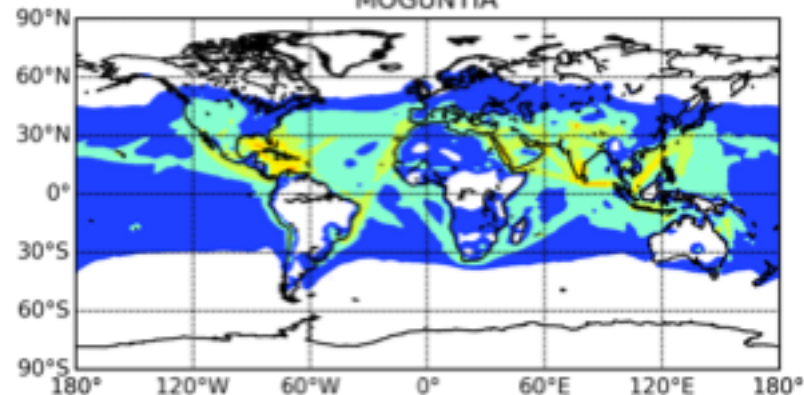


# OH: CB05 vs MOGUNTIA

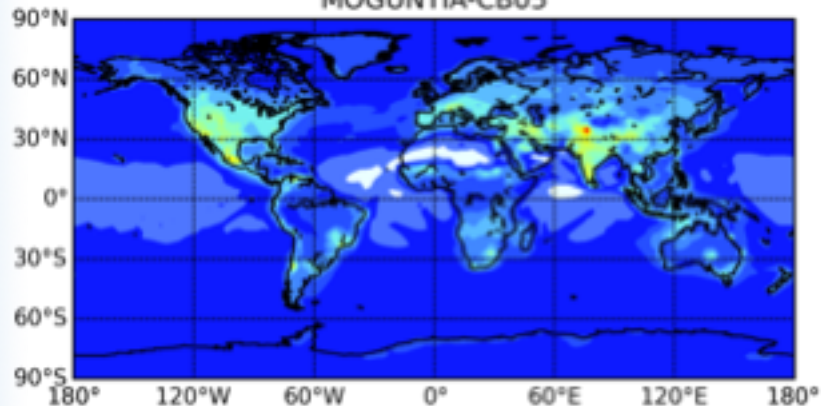
OH, Surface, Annual Mean  
CB05



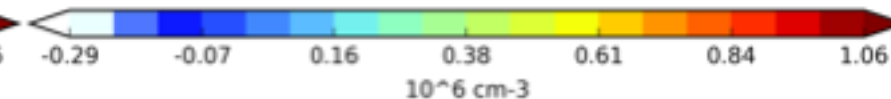
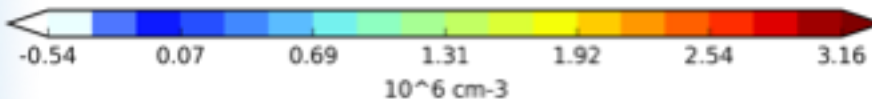
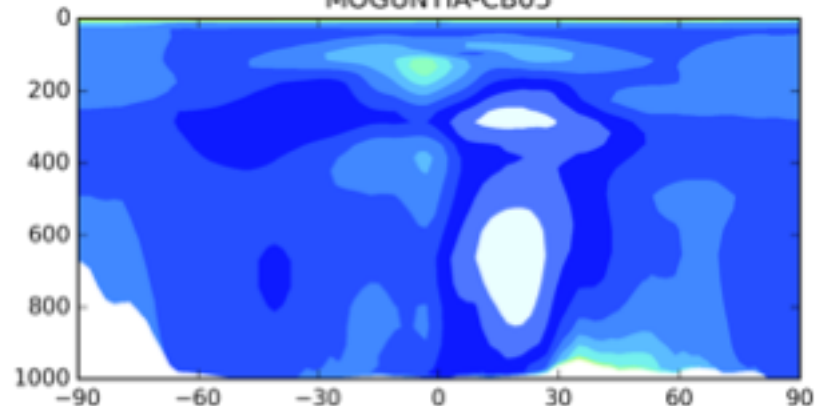
OH, Surface, Annual Mean  
MOGUNTIA



OH, Surface, Annual Mean  
MOGUNTIA-CB05



OH, Zonal Mean, Annual Mean  
MOGUNTIA-CB05



# CH<sub>4</sub> : CB05 vs MOGUNTIA

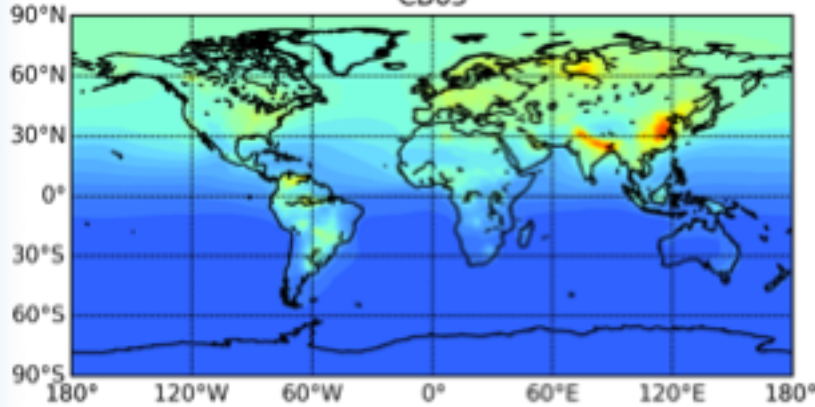
Global Lifetime\*

8.1 years

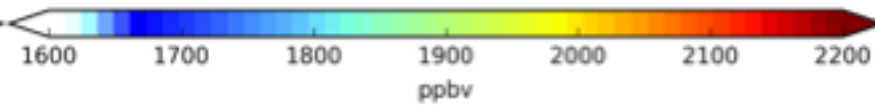
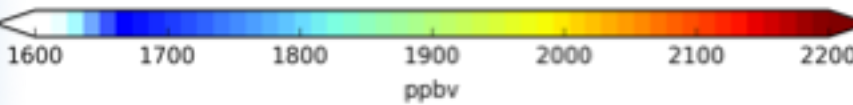
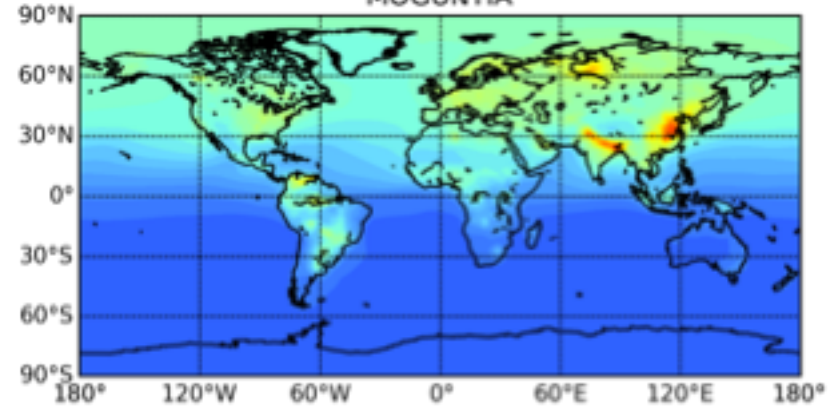


8.5 years

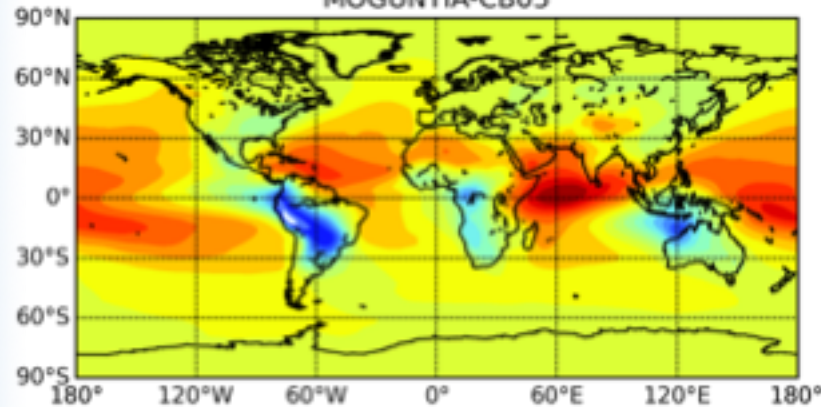
CH<sub>4</sub>, Surface, Annual Mean  
CB05



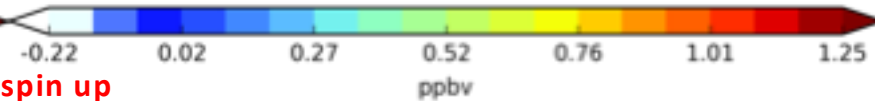
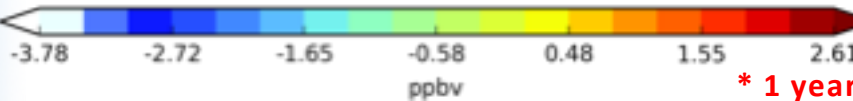
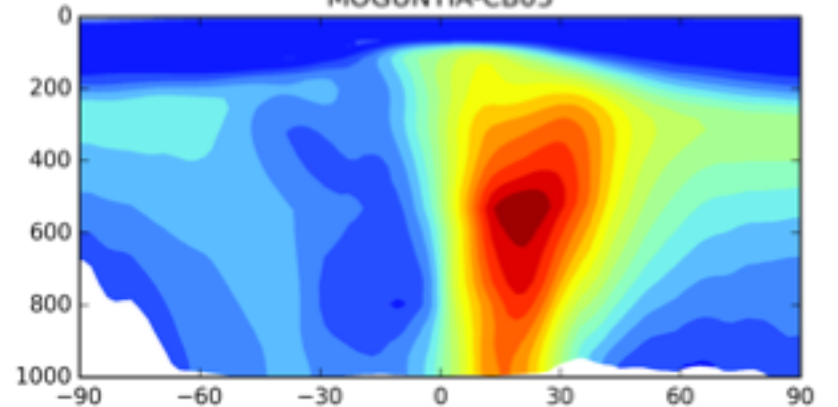
CH<sub>4</sub>, Surface, Annual Mean  
MOGUNTIA



CH<sub>4</sub>, Surface, Annual Mean  
MOGUNTIA-CB05



CH<sub>4</sub>, Zonal Mean, Annual Mean  
MOGUNTIA-CB05

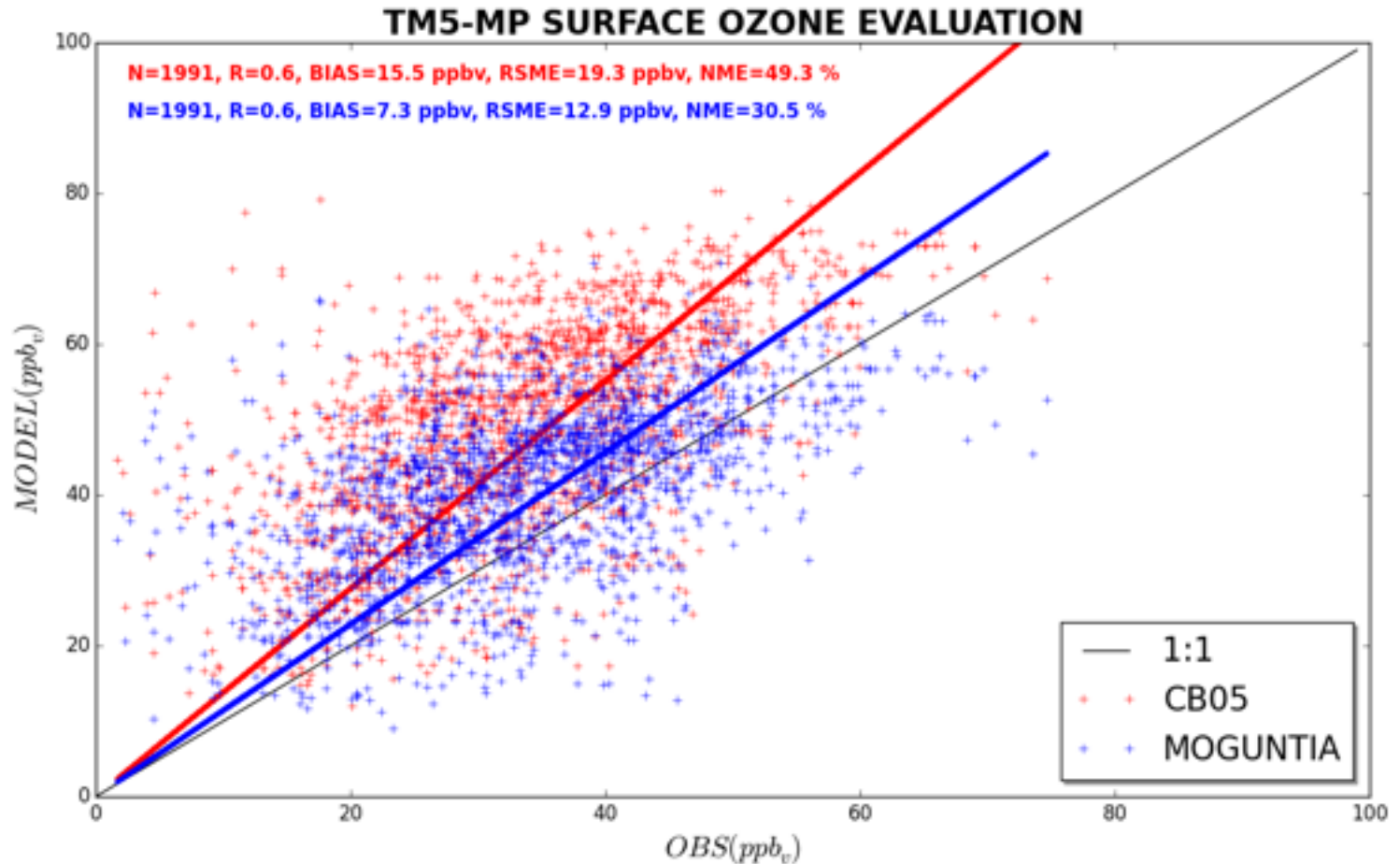


\* 1 year spin up



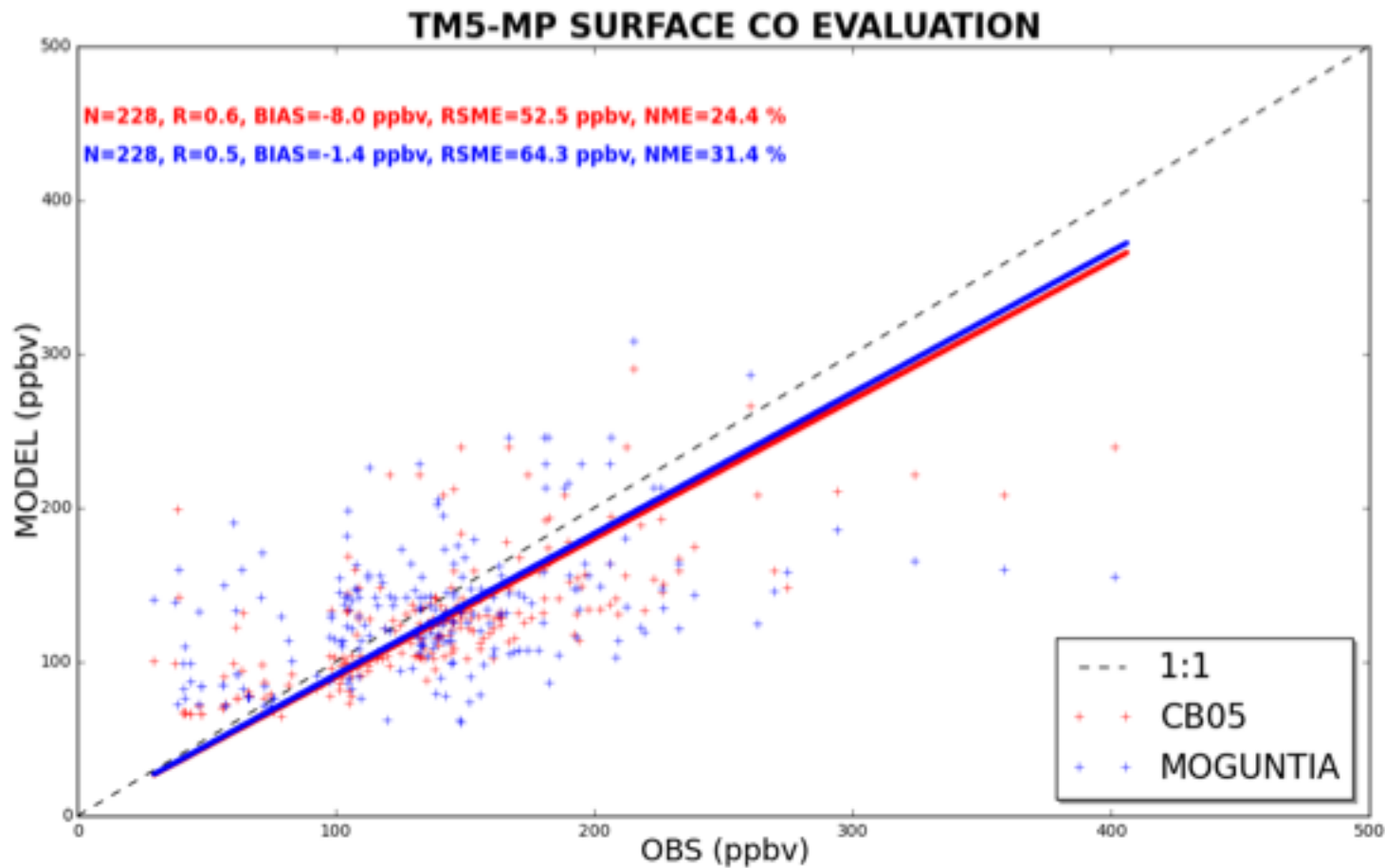


# Ozone Evaluation



*O<sub>3</sub> surface observations obtained from the World Data Centre for Greenhouse Gases (WDCGG; <http://ds.data.jma.go.jp/gmd/wdcgg/introduction.html>) and from the European Monitoring and Evaluation Programme (EMEP; <http://www.emep.int>).*

# CO Evaluation



*CO surface observations obtained from the World Data Centre for Greenhouse Gases (WDCGG; <http://ds.data.jma.go.jp/gmd/wdcgg/introduction.html>).*

# Summary

- TM5-MP is now also coupled with an explicit gas-phase chemical scheme (i.e., MOGUNTIA) - simulation year 2006 with one year (i.e., 2005) spin-up.
- MOGUNTIA (and CB05) chemical scheme is written in KPP.
- KPP-MOGUNTIA is ~28% slower than EBI-CB05 and ~12% than KPP-CB05, all for a glb600x400-tropo25 configuration.
- MOGUNTIA chemistry seems to improve the simulated O3 and CO validations with lower O3 and slightly higher CO global concentrations.
- Further developments:
  - TM5-MP is also coupled with a multiphase scheme in cloud droplets and wet aerosols
  - A „light“ AQCHEM version is implemented in EC-Earth but based on the CMB05 mechanism
  - A paper on the MOGUNTIA-TM5MP coupling is in preparation





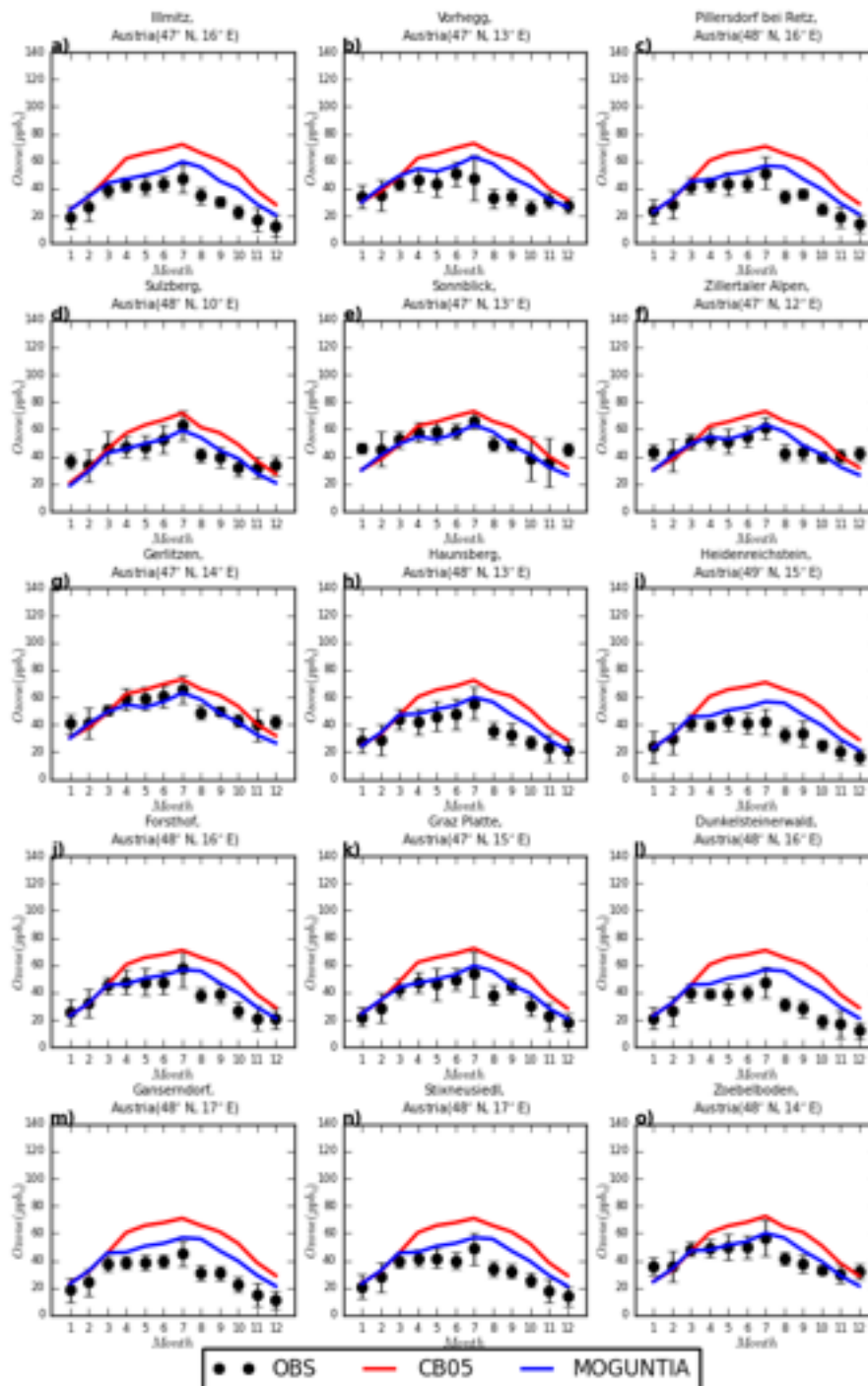


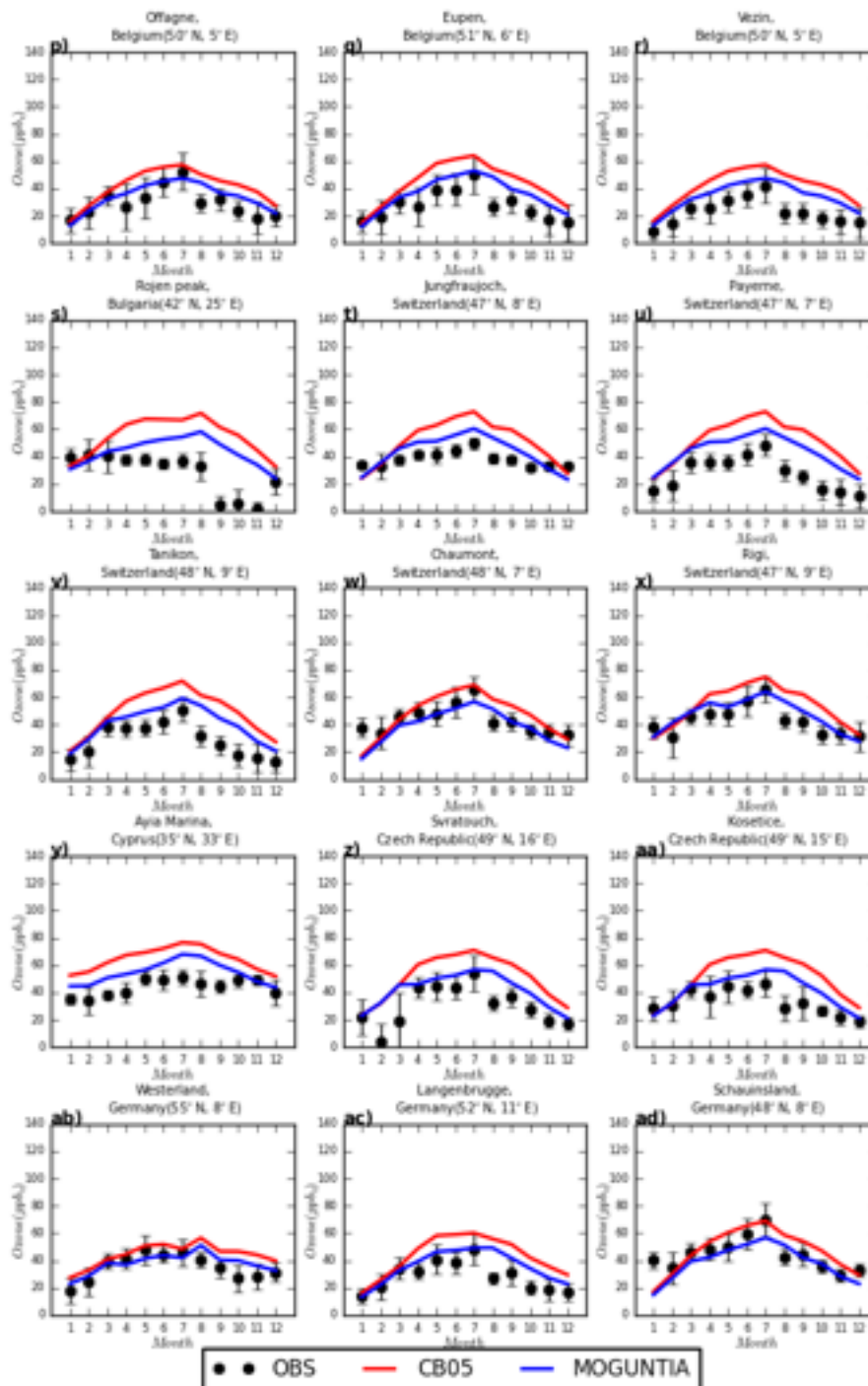
# Appendix





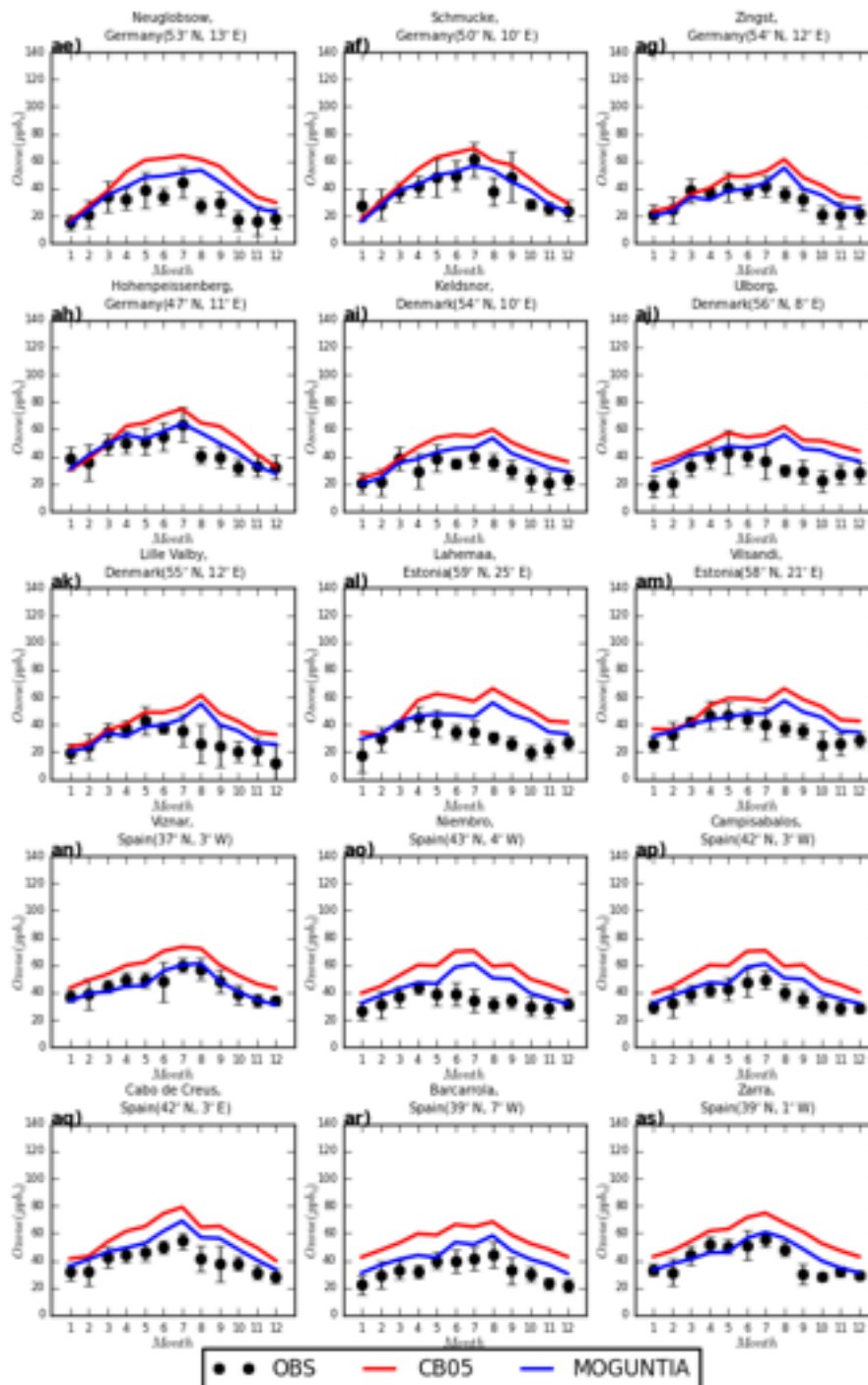
# 1. O<sub>3</sub>

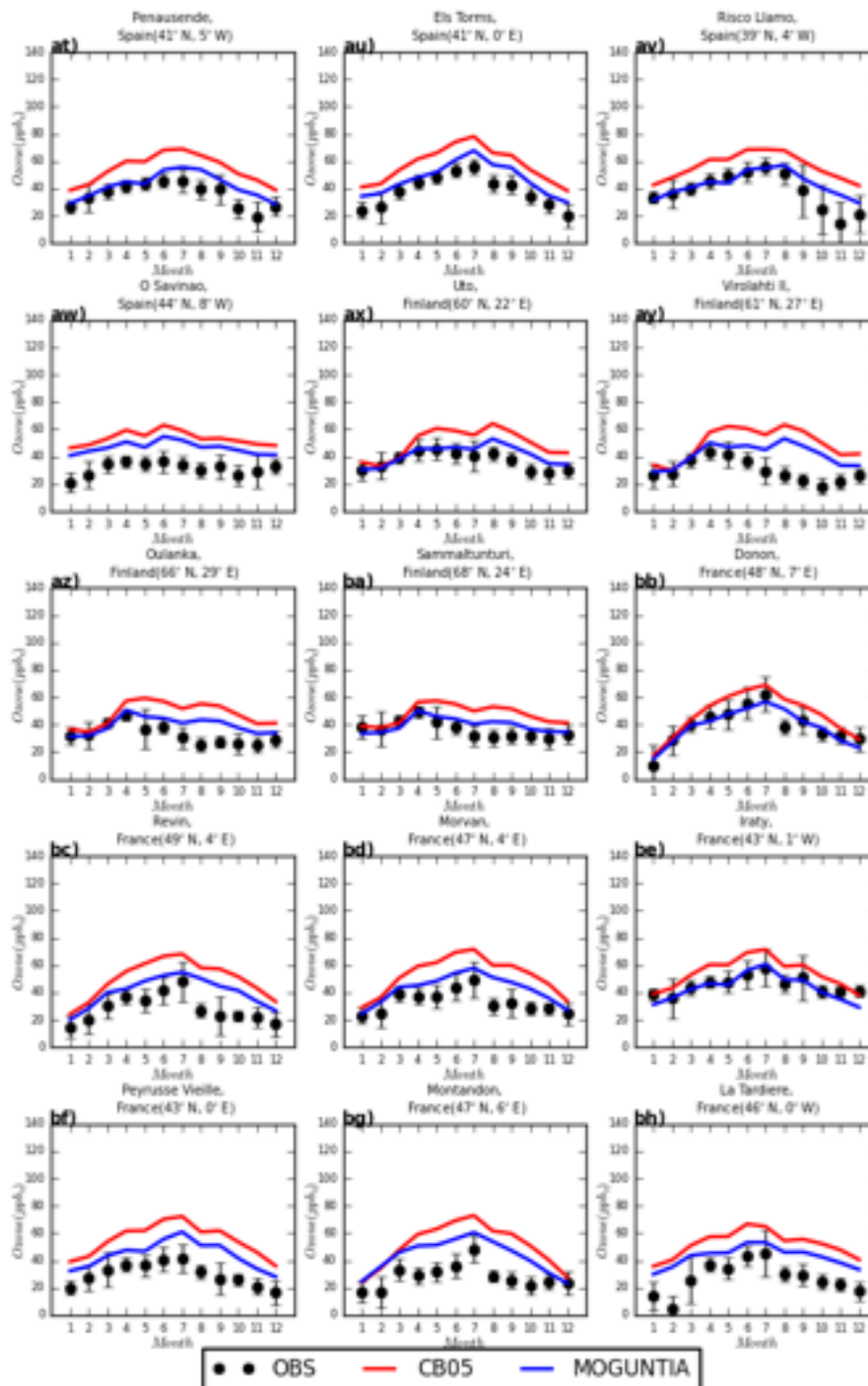




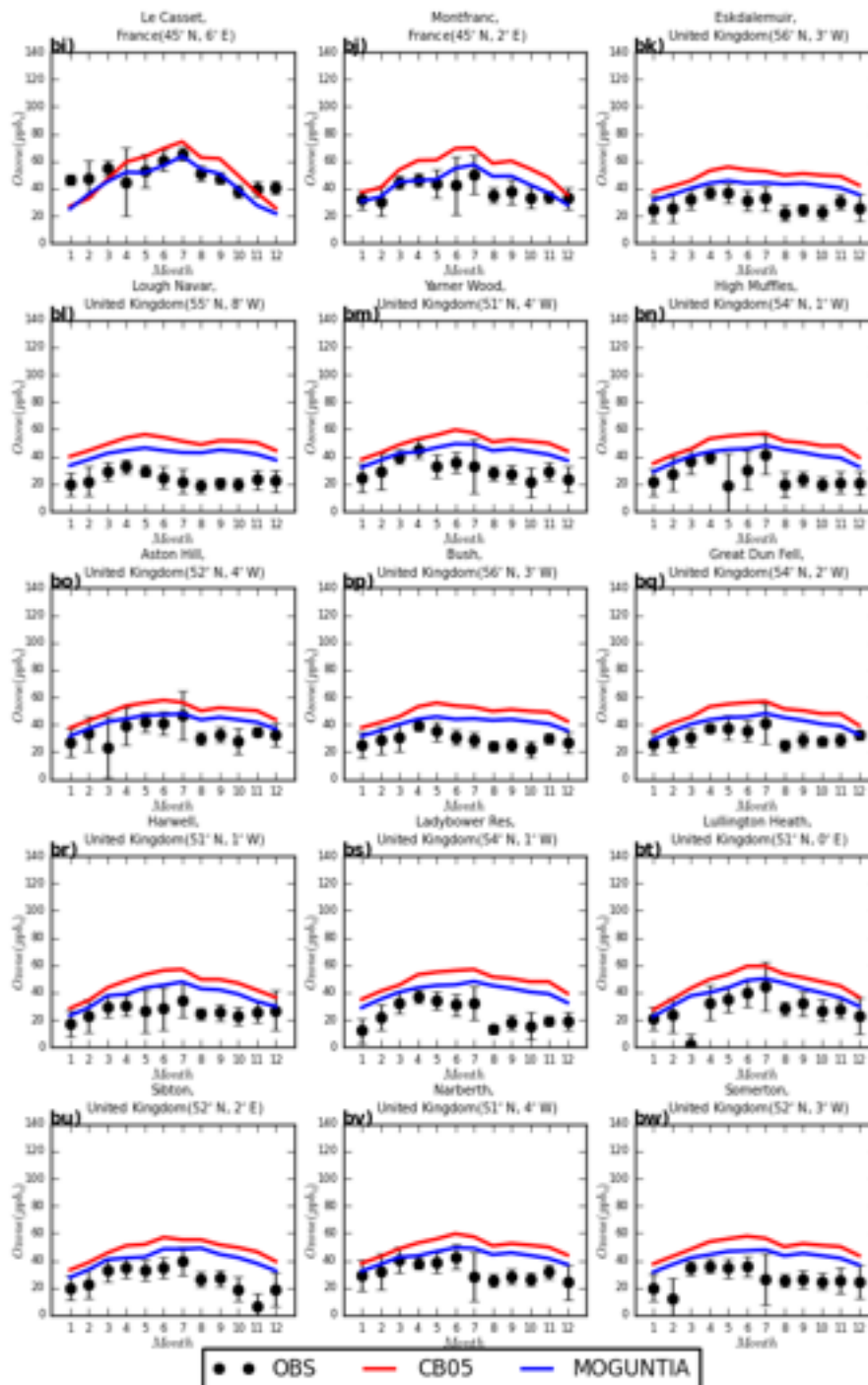


DEON



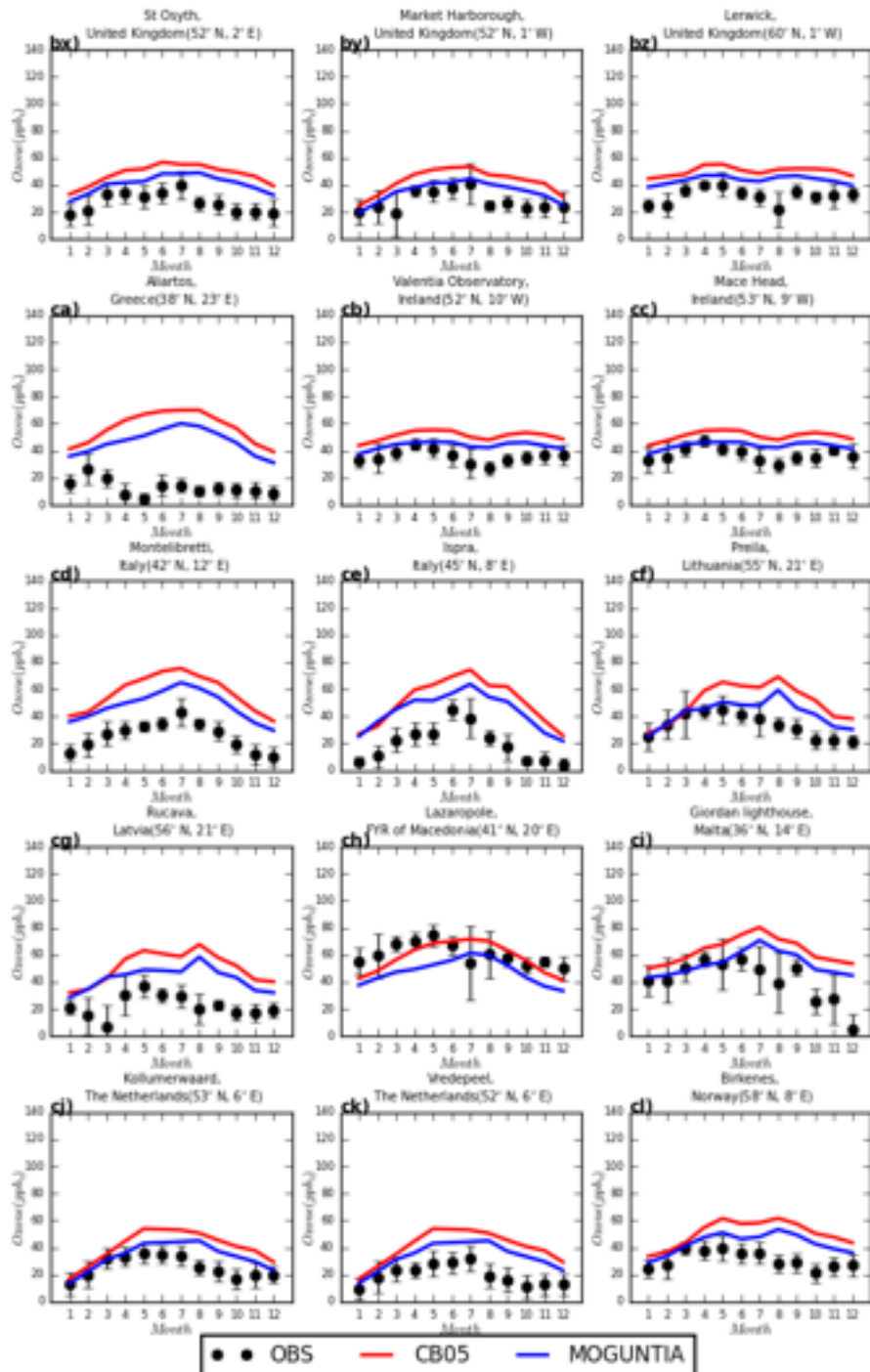


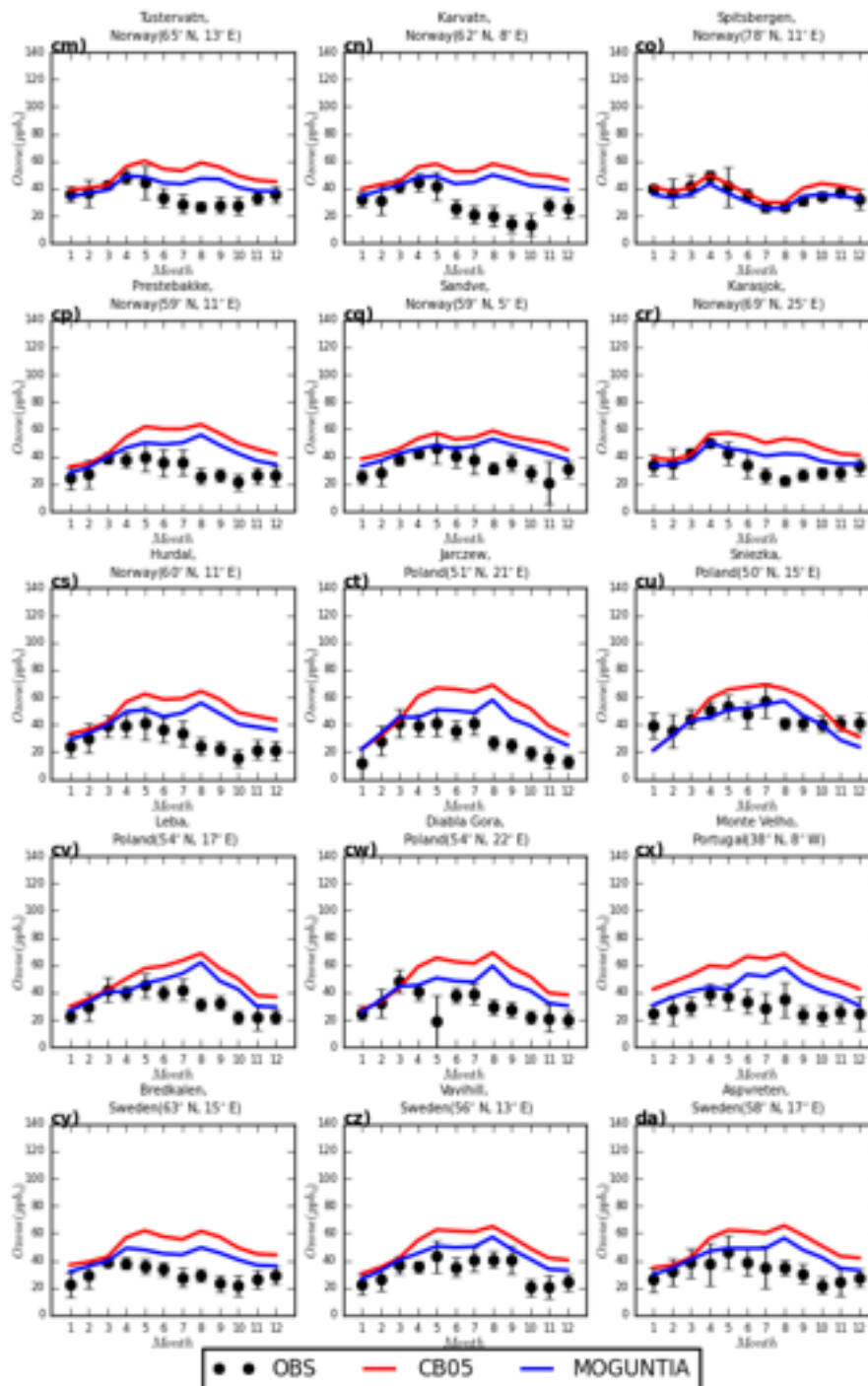


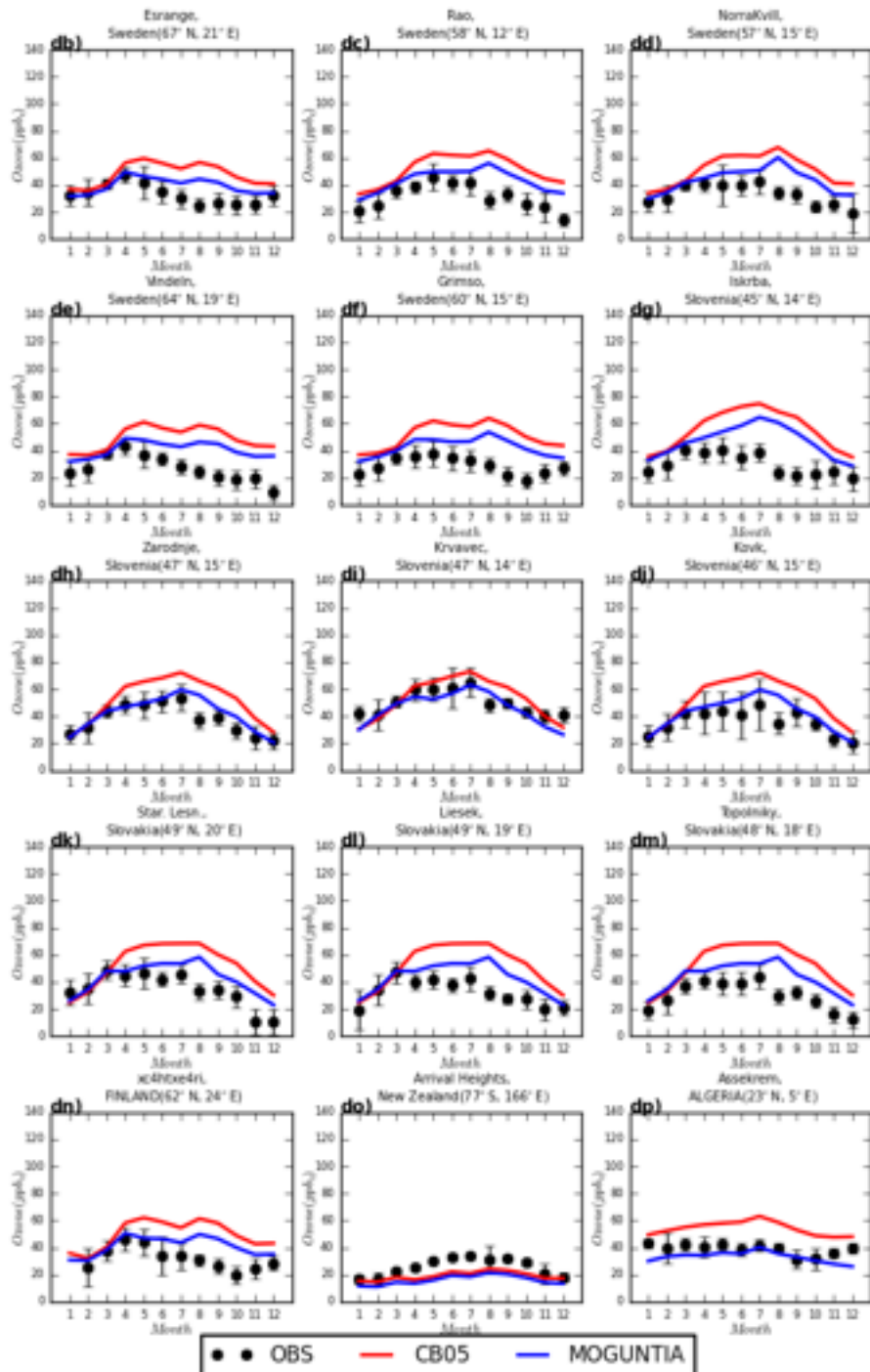


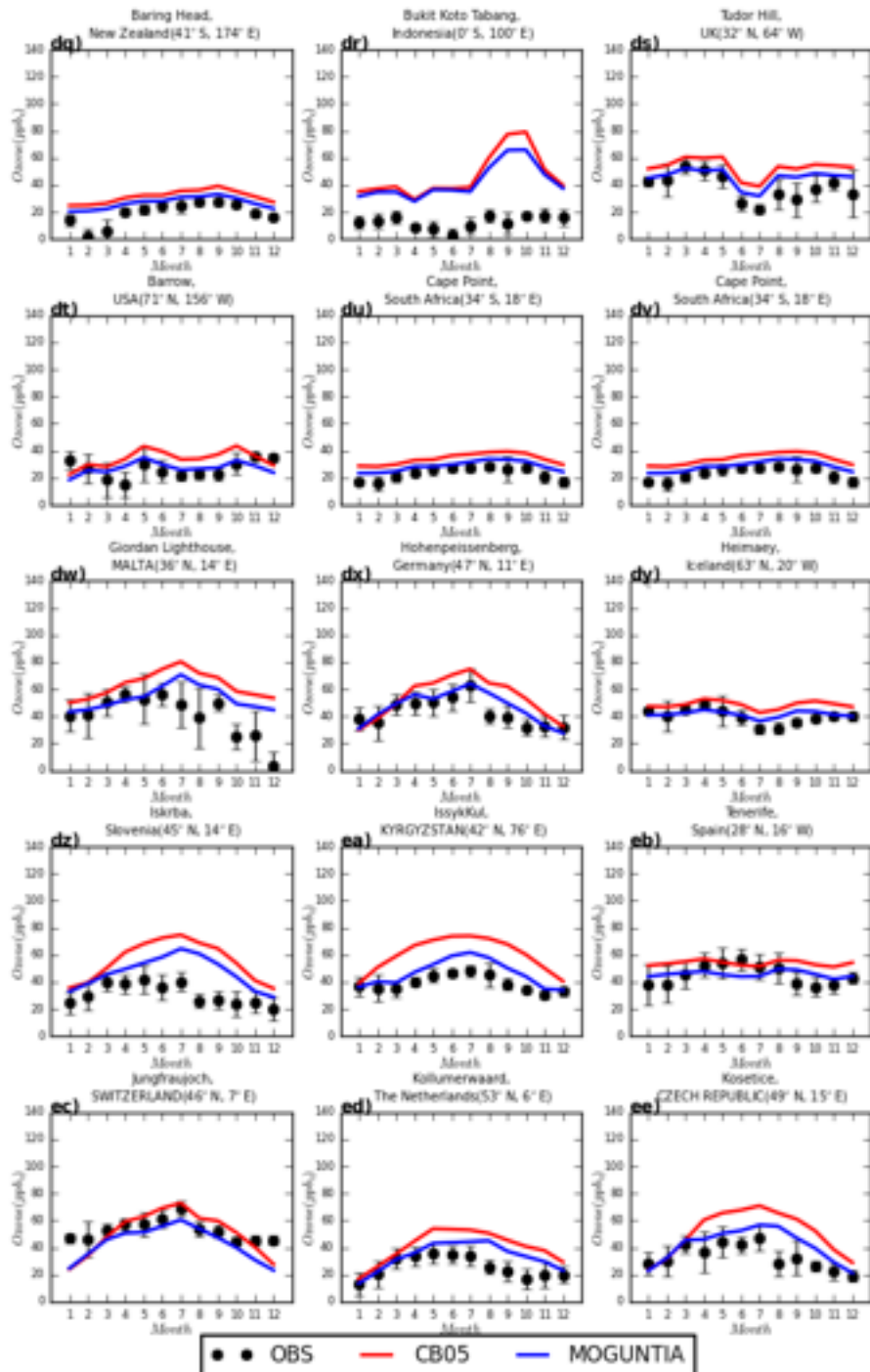


DEON





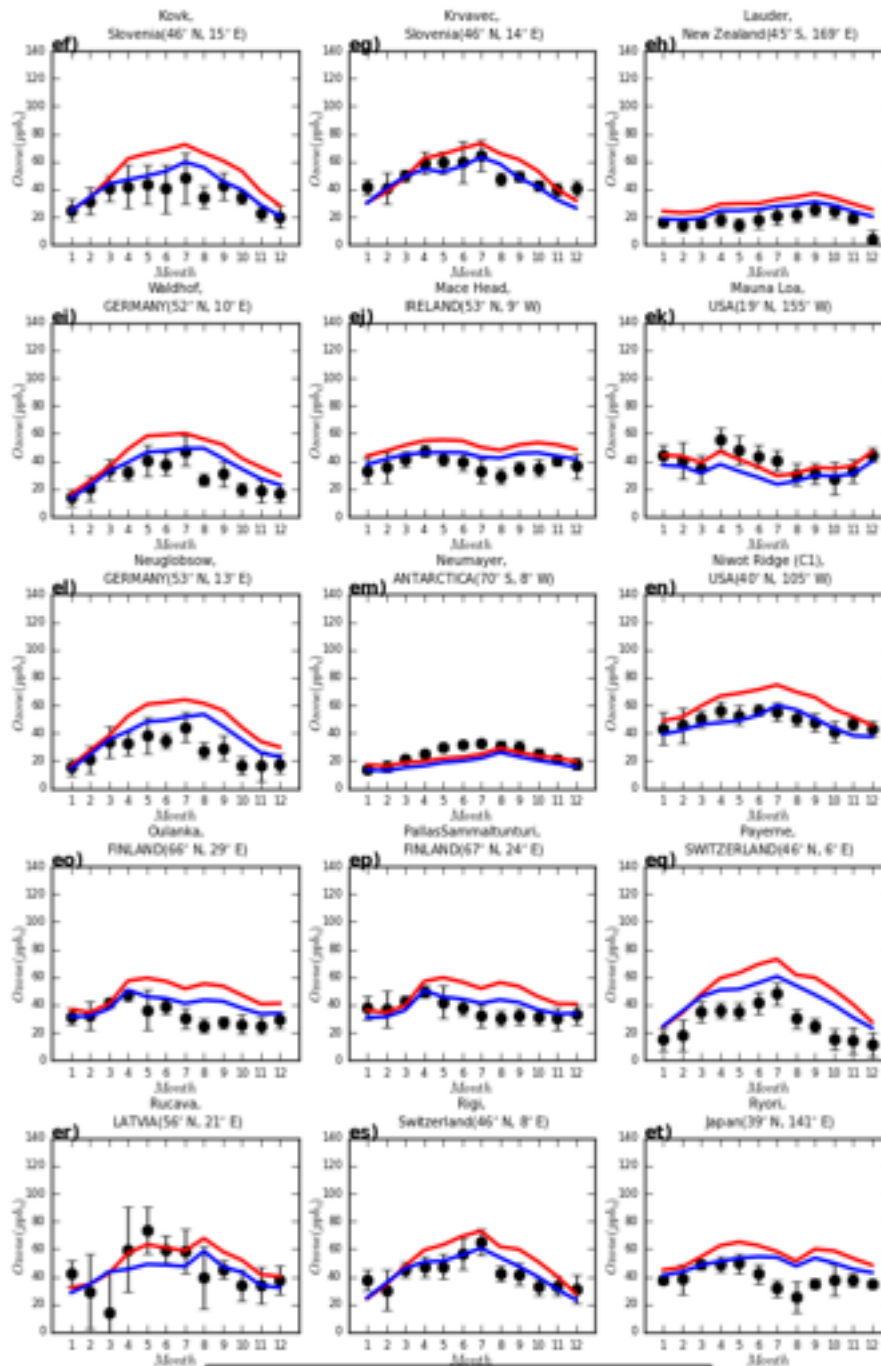




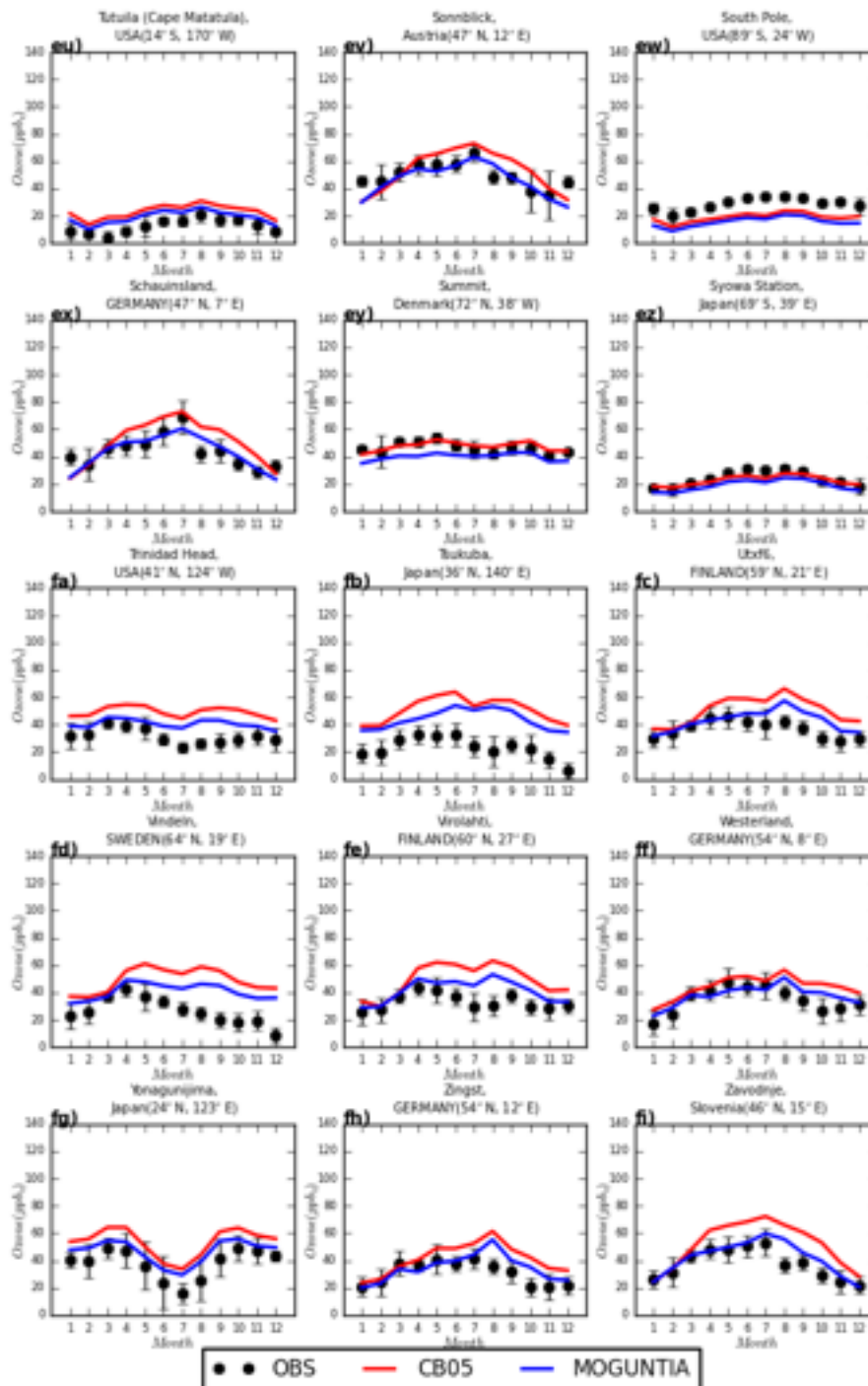




DEON

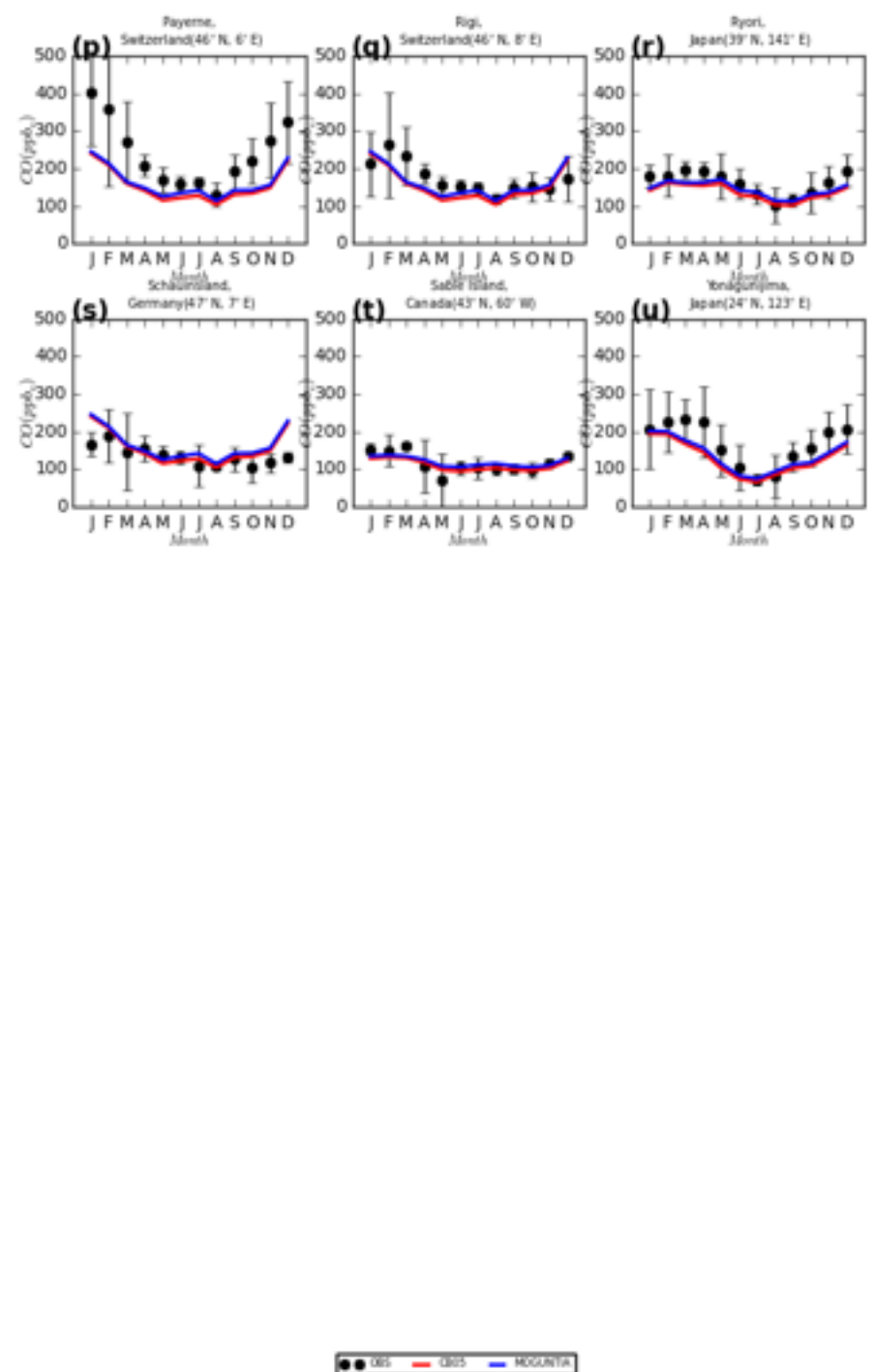
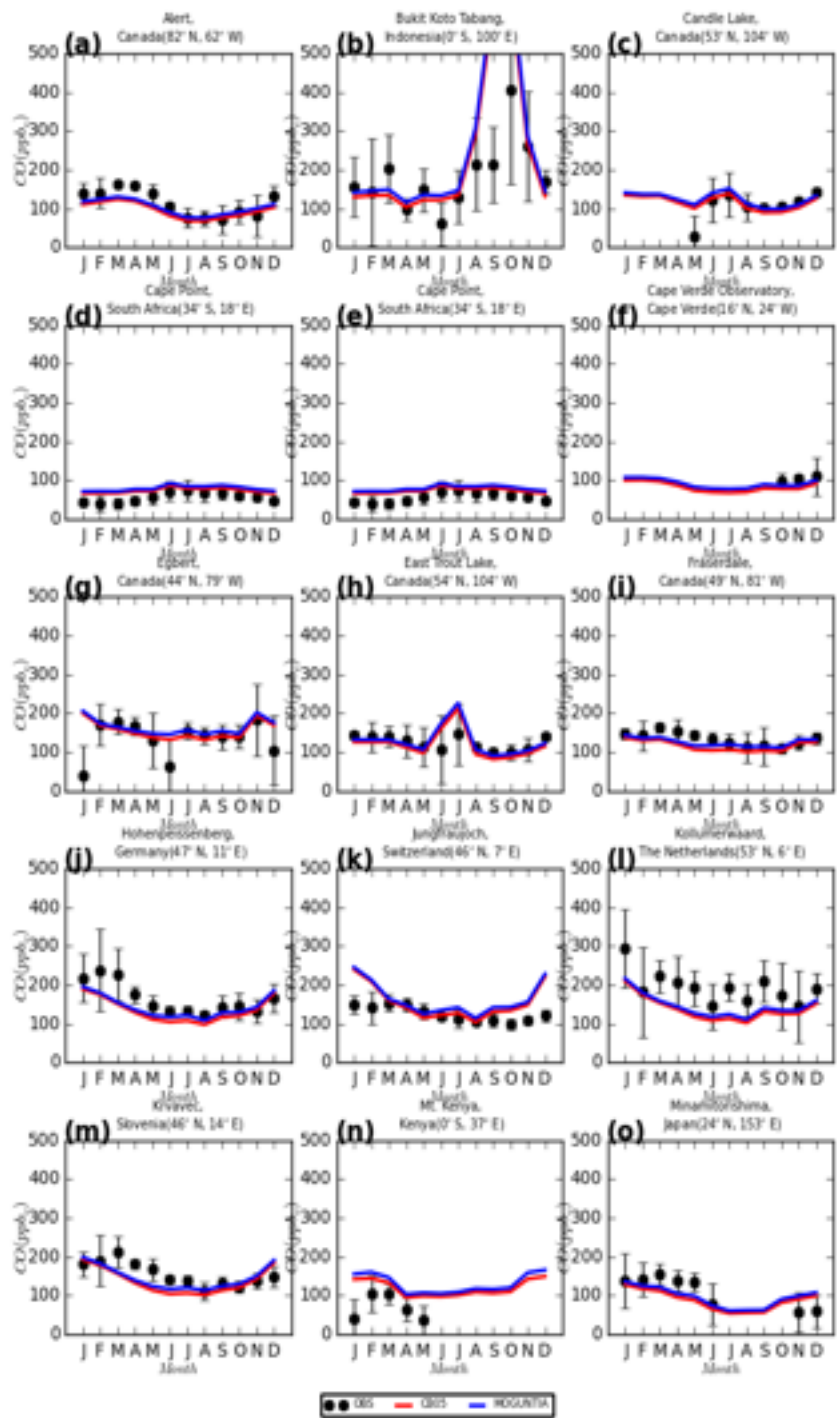






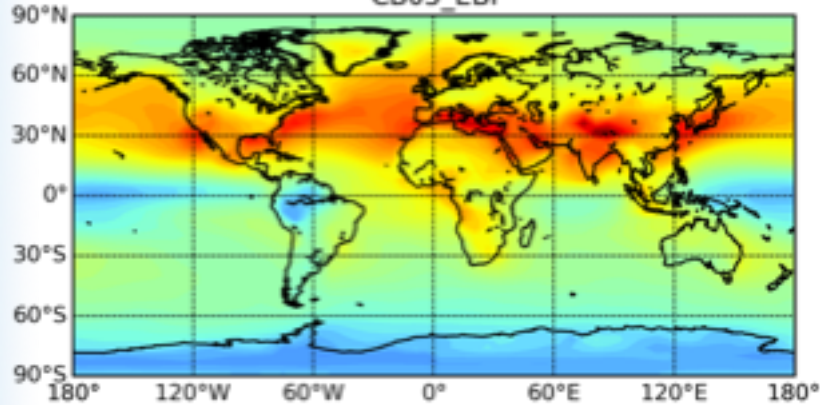


## 2. CO

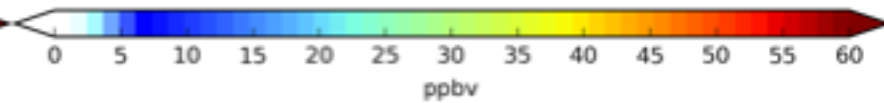
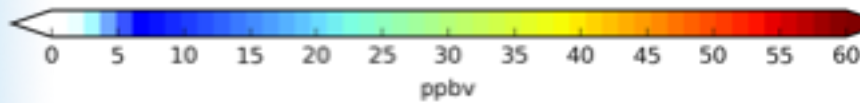
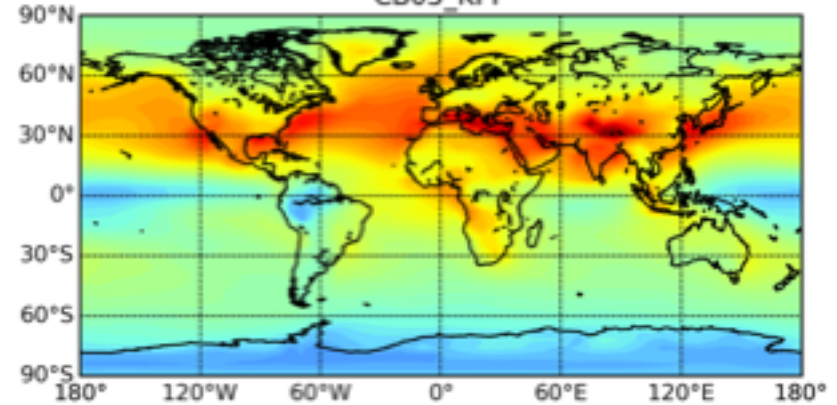


# 3. O<sub>3</sub>: EBI vs KPP

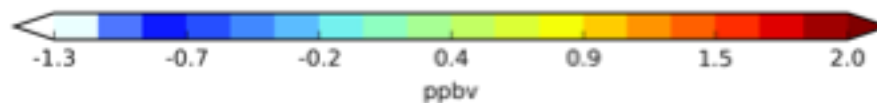
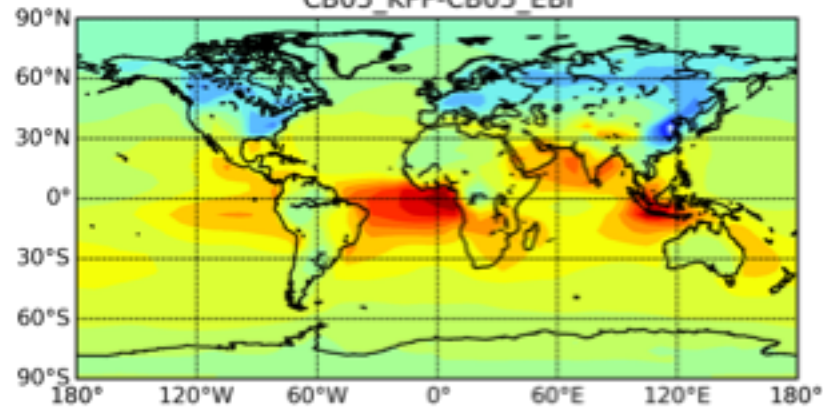
O<sub>3</sub>, Surface, Annual Mean  
CB05\_EBI



O<sub>3</sub>, Surface, Annual Mean  
CB05\_KPP

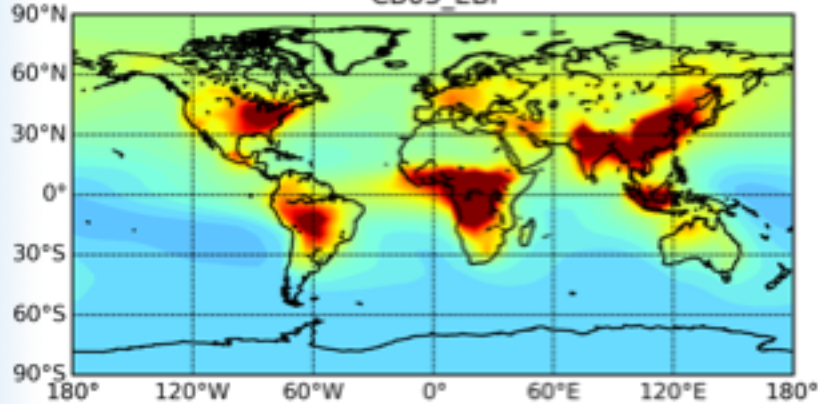


O<sub>3</sub>, Surface, Annual Mean  
CB05\_KPP-CB05\_EBI

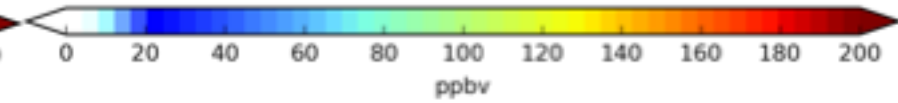
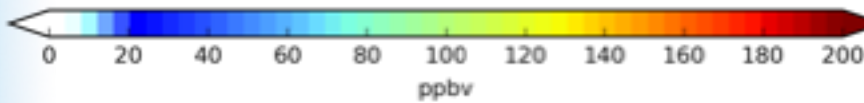
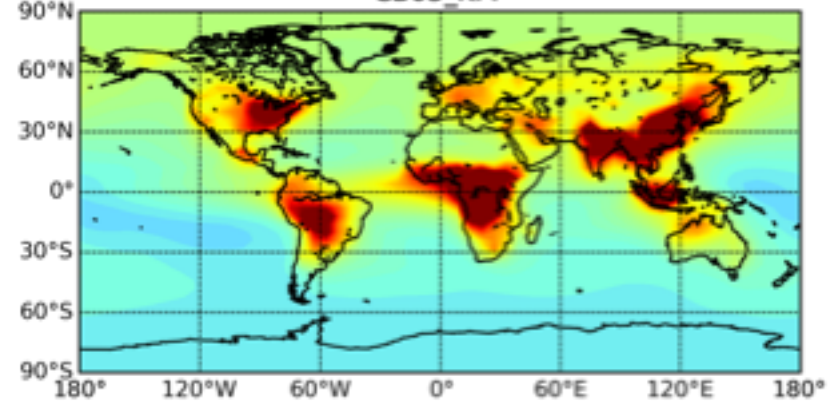


# 4. CO: EBI vs KPP

CO, Surface, Annual Mean  
CB05\_EBI



CO, Surface, Annual Mean  
CB05\_KPP



CO, Surface, Annual Mean  
CB05\_KPP-CB05\_EBI

